

# 1-Nitroanthraquinone

<b>Other names:</b>	1-Nitroanthrachinon 1-nitro-9,10-anthracenedione 9,10-Anthracenedione, 1-nitro- Anthraquinone, 1-nitro- «alpha»-Nitroanthraquinone
<b>Inchi:</b>	InChI=1S/C14H7NO4/c16-13-8-4-1-2-5-9(8)14(17)12-10(13)6-3-7-11(12)15(18)19/h1-7H
<b>InchiKey:</b>	YCANAXVBJKNANM-UHFFFAOYSA-N
<b>Formula:</b>	C14H7NO4
<b>SMILES:</b>	O=C1c2ccccc2C(=O)c2c1ccc2[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	253.21
<b>CAS:</b>	82-34-8

## Physical Properties

Property code	Value	Unit	Source
gf	133.86	kJ/mol	Joback Method
hf	-80.50	kJ/mol	Joback Method
hfus	28.48	kJ/mol	Joback Method
hvap	78.43	kJ/mol	Joback Method
log10ws	-4.28		Crippen Method
logp	2.370		Crippen Method
mcvol	170.300	ml/mol	McGowan Method
pc	3360.64	kPa	Joback Method
rinpol	394.38		NIST Webbook
rinpol	394.38		NIST Webbook
tb	882.64	K	Joback Method
tc	1172.56	K	Joback Method
tf	643.69	K	Joback Method
vc	0.665	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	487.16	J/molxK	882.64	Joback Method
cpg	498.00	J/molxK	930.96	Joback Method

cpg	507.52	J/mol×K	979.28	Joback Method
cpg	515.80	J/mol×K	1027.60	Joback Method
cpg	522.88	J/mol×K	1075.92	Joback Method
cpg	528.82	J/mol×K	1124.24	Joback Method
cpg	533.67	J/mol×K	1172.56	Joback Method
hsubt	139.70	kJ/mol	423.50	NIST Webbook
hsubt	108.90 ± 2.10	kJ/mol	396.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solubility of 1-aminoanthraquinone and 1-nitroanthraquinone in supercritical carbon dioxide:</b>	<a href="https://www.doi.org/10.1016/j.jct.2016.09.032">https://www.doi.org/10.1016/j.jct.2016.09.032</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C82348&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C82348&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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