

Benzene, 4-(chloromethyl)-1,2-dimethyl-

Other names:	1,2-Dimethyl-4-(chloromethyl)benzene 3,4-Dimethylbenzyl chloride 4-(chloromethyl)-o-xylene Benzene, 1-(chloromethyl)-3,4-dimethyl-
Inchi:	InChI=1S/C9H11Cl/c1-7-3-4-9(6-10)5-8(7)2/h3-5H,6H2,1-2H3
InchiKey:	UBQRAAXAHIKWRI-UHFFFAOYSA-N
Formula:	C9H11Cl
SMILES:	Cc1ccc(CCl)cc1C
Mol. weight [g/mol]:	154.64
CAS:	102-46-5

Physical Properties

Property code	Value	Unit	Source
gf	106.12	kJ/mol	Joback Method
hf	-31.24	kJ/mol	Joback Method
hfus	16.53	kJ/mol	Joback Method
hvap	43.61	kJ/mol	Joback Method
log10ws	-3.46		Crippen Method
logp	3.042		Crippen Method
mcvol	126.150	ml/mol	McGowan Method
pc	3025.61	kPa	Joback Method
rinpol	1233.00		NIST Webbook
rinpol	1200.00		NIST Webbook
tb	479.39	K	Joback Method
tc	695.70	K	Joback Method
tf	272.57	K	Joback Method
vc	0.480	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	243.19	J/mol×K	479.39	Joback Method
cpg	299.10	J/mol×K	659.65	Joback Method
cpg	289.18	J/mol×K	623.60	Joback Method

cpg	278.64	J/molxK	587.55	Joback Method
cpg	267.48	J/molxK	551.49	Joback Method
cpg	255.67	J/molxK	515.44	Joback Method
cpg	308.45	J/molxK	695.70	Joback Method
dvisc	0.0002356	Paxs	479.39	Joback Method
dvisc	0.0002888	Paxs	444.92	Joback Method
dvisc	0.0003664	Paxs	410.45	Joback Method
dvisc	0.0004855	Paxs	375.98	Joback Method
dvisc	0.0006809	Paxs	341.51	Joback Method
dvisc	0.0010303	Paxs	307.04	Joback Method
dvisc	0.0017313	Paxs	272.57	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	389.70	K	3.20	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38272e+01
Coeff. B	-3.79701e+03
Coeff. C	-7.51370e+01
Temperature range (K), min.	355.57
Temperature range (K), max.	521.02

Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C102465&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cp_g:	Ideal gas heat capacity
dv_{isc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
ri_{npol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_{brp}:	Boiling point at reduced pressure
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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