

# Fumaric acid, monoamide, N-(5-chloro-2-methoxyphenyl)-, 3,5-difluorophenyl ester

**Other names:** Fumaric acid, monoamide, N-(5-chloro-2-methoxyphenyl)-, 3,5-fluorophenyl ester  
**Inchi:** InChI=1S/C17H12ClF2NO4/c1-24-15-3-2-10(18)6-14(15)21-16(22)4-5-17(23)25-13-8-11  
**InchiKey:** NULCGJZYDRWFCK-SNAWJCMRSA-N  
**Formula:** C17H12ClF2NO4  
**SMILES:** COc1ccc(Cl)cc1NC(=O)C=CC(=O)Oc1cc(F)cc(F)c1  
**Mol. weight [g/mol]:** 367.73

## Physical Properties

Property code	Value	Unit	Source
gf	-421.22	kJ/mol	Joback Method
hf	-693.90	kJ/mol	Joback Method
hfus	47.54	kJ/mol	Joback Method
hvap	88.09	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	3.727		Crippen Method
mcvol	239.210	ml/mol	McGowan Method
pc	2032.72	kPa	Joback Method
rinpol	2936.00		NIST Webbook
tb	904.52	K	Joback Method
tc	1132.97	K	Joback Method
tf	607.27	K	Joback Method
vc	0.919	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	668.32	J/molxK	904.52	Joback Method
cpg	678.11	J/molxK	942.59	Joback Method
cpg	686.87	J/molxK	980.67	Joback Method
cpg	694.61	J/molxK	1018.74	Joback Method
cpg	701.37	J/molxK	1056.82	Joback Method
cpg	707.19	J/molxK	1094.89	Joback Method
cpg	712.09	J/molxK	1132.97	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357410&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357410&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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