

# Ethoxyacetaldehyde diethylacetal

<b>Other names:</b>	1,1,2-Triethoxyethane Ethane, 1,1,2-triethoxy-
<b>Inchi:</b>	InChI=1S/C8H18O3/c1-4-9-7-8(10-5-2)11-6-3/h8H,4-7H2,1-3H3
<b>InchiKey:</b>	VNSJUZIHZNZLKM-UHFFFAOYSA-N
<b>Formula:</b>	C8H18O3
<b>SMILES:</b>	CCOCC(OCC)OCC
<b>Mol. weight [g/mol]:</b>	162.23
<b>CAS:</b>	4819-77-6

## Physical Properties

Property code	Value	Unit	Source
gf	-300.96	kJ/mol	Joback Method
hf	-610.39	kJ/mol	Joback Method
hfus	16.52	kJ/mol	Joback Method
hvap	40.24	kJ/mol	Joback Method
log10ws	-1.04		Crippen Method
logp	1.422		Crippen Method
mvol	141.190	ml/mol	McGowan Method
pc	2429.05	kPa	Joback Method
tb	449.26	K	Joback Method
tc	617.01	K	Joback Method
tf	231.61	K	Joback Method
vc	0.531	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	307.27	J/molxK	449.26	Joback Method
cpg	367.33	J/molxK	589.05	Joback Method
cpg	355.98	J/molxK	561.09	Joback Method
cpg	344.29	J/molxK	533.14	Joback Method
cpg	332.26	J/molxK	505.18	Joback Method
cpg	319.92	J/molxK	477.22	Joback Method
cpg	378.33	J/molxK	617.01	Joback Method

dvisc	0.0001554	Paxs	449.26	Joback Method
dvisc	0.0002070	Paxs	412.99	Joback Method
dvisc	0.0002915	Paxs	376.71	Joback Method
dvisc	0.0004415	Paxs	340.44	Joback Method
dvisc	0.0007383	Paxs	304.16	Joback Method
dvisc	0.0014191	Paxs	267.88	Joback Method
dvisc	0.0033474	Paxs	231.61	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4819776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4819776&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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