

# Tripropylene glycol, monoallyl ether, acetate

**Inchi:** InChI=1S/C14H26O5/c1-6-7-16-8-11(2)17-9-12(3)18-10-13(4)19-14(5)15/h6,11-13H,1,7-  
**InchiKey:** AAVVZXYBUPOMCC-UHFFFAOYSA-N  
**Formula:** C14H26O5  
**SMILES:** C=CCOCC(C)OCC(C)OCC(C)OC(C)=O  
**Mol. weight [g/mol]:** 274.35

## Physical Properties

Property code	Value	Unit	Source
gf	-401.40	kJ/mol	Joback Method
hf	-864.16	kJ/mol	Joback Method
hfus	26.52	kJ/mol	Joback Method
hvap	61.31	kJ/mol	Joback Method
log10ws	-2.00		Crippen Method
logp	1.951		Crippen Method
mcvol	228.870	ml/mol	McGowan Method
pc	1598.72	kPa	Joback Method
rinpol	1603.00		NIST Webbook
rinpol	1604.00		NIST Webbook
rinpol	1600.00		NIST Webbook
rinpol	1601.00		NIST Webbook
rinpol	1600.00		NIST Webbook
rinpol	1601.00		NIST Webbook
rinpol	1604.00		NIST Webbook
rinpol	1601.00		NIST Webbook
rinpol	1602.00		NIST Webbook
rinpol	1600.00		NIST Webbook
rinpol	1602.00		NIST Webbook
rinpol	1602.00		NIST Webbook
tb	658.63	K	Joback Method
tc	836.61	K	Joback Method
tf	339.63	K	Joback Method
vc	0.861	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	633.86	J/molxK	658.63	Joback Method
cpg	709.19	J/molxK	806.95	Joback Method
cpg	695.66	J/molxK	777.29	Joback Method
cpg	681.35	J/molxK	747.62	Joback Method
cpg	666.27	J/molxK	717.96	Joback Method
cpg	650.44	J/molxK	688.29	Joback Method
cpg	721.94	J/molxK	836.61	Joback Method
dvisc	0.0000605	Paxs	658.63	Joback Method
dvisc	0.0000835	Paxs	605.46	Joback Method
dvisc	0.0001225	Paxs	552.30	Joback Method
dvisc	0.0001951	Paxs	499.13	Joback Method
dvisc	0.0003470	Paxs	445.96	Joback Method
dvisc	0.0007215	Paxs	392.80	Joback Method
dvisc	0.0018864	Paxs	339.63	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R152320&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R152320&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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