

# 2-Chloropropionic acid, phenyl ester

<b>Inchi:</b>	InChI=1S/C9H9ClO2/c1-7(10)9(11)12-8-5-3-2-4-6-8/h2-7H,1H3
<b>InchiKey:</b>	PTFWCFRZCGETRS-UHFFFAOYSA-N
<b>Formula:</b>	C9H9ClO2
<b>SMILES:</b>	CC(Cl)C(=O)Oc1ccccc1
<b>Mol. weight [g/mol]:</b>	184.62

## Physical Properties

Property code	Value	Unit	Source
gf	-110.98	kJ/mol	Joback Method
hf	-258.38	kJ/mol	Joback Method
hfus	16.57	kJ/mol	Joback Method
hvap	51.06	kJ/mol	Joback Method
log10ws	-2.47		Crippen Method
logp	2.219		Crippen Method
mcvol	133.590	ml/mol	McGowan Method
pc	3325.84	kPa	Joback Method
rinpola	1287.00		NIST Webbook
tb	545.28	K	Joback Method
tc	771.28	K	Joback Method
tf	304.69	K	Joback Method
vc	0.498	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.82	J/molxK	545.28	Joback Method
cpg	335.15	J/molxK	733.61	Joback Method
cpg	325.96	J/molxK	695.95	Joback Method
cpg	316.05	J/molxK	658.28	Joback Method
cpg	305.41	J/molxK	620.61	Joback Method
cpg	294.01	J/molxK	582.95	Joback Method
cpg	343.63	J/molxK	771.28	Joback Method
dvisc	0.0002204	Paxs	545.28	Joback Method
dvisc	0.0002851	Paxs	505.18	Joback Method

dvisc	0.0003856	Paxs	465.08	Joback Method
dvisc	0.0005521	Paxs	424.99	Joback Method
dvisc	0.0008517	Paxs	384.89	Joback Method
dvisc	0.0014534	Paxs	344.79	Joback Method
dvisc	0.0028549	Paxs	304.69	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357775&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357775&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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