

Piperazine, 2,6-dimethyl-

Other names:	2,6-Dimethylpiperazine
Inchi:	InChI=1S/C6H14N2/c1-5-3-7-4-6(2)8-5/h5-8H,3-4H2,1-2H3
InchiKey:	IFNWESYYDINUHV-UHFFFAOYSA-N
Formula:	C6H14N2
SMILES:	CC1CNCC(C)N1
Mol. weight [g/mol]:	114.19
CAS:	108-49-6

Physical Properties

Property code	Value	Unit	Source
gf	191.80	kJ/mol	Joback Method
hf	-57.57	kJ/mol	Joback Method
hfus	23.38	kJ/mol	Joback Method
hvap	42.59	kJ/mol	Joback Method
log10ws	-0.83		Crippen Method
logp	-0.044		Crippen Method
mcvol	104.500	ml/mol	McGowan Method
pc	4046.64	kPa	Joback Method
tb	435.20	K	NIST Webbook
tc	646.00	K	Critical temperatures and pressures of caprolactam, dimethyl sulfoxide, 1,4-dimethylpiperazine, and 2,6-dimethylpiperazine
tf	370.58	K	Joback Method
vc	0.378	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	216.08	J/mol×K	448.66	Joback Method
cpg	232.09	J/mol×K	485.59	Joback Method
cpg	247.43	J/mol×K	522.51	Joback Method
cpg	262.10	J/mol×K	559.44	Joback Method

cpg	276.10	J/mol×K	596.36	Joback Method
cpg	289.42	J/mol×K	633.29	Joback Method
cpg	302.05	J/mol×K	670.21	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.55423e+01
Coeff. B	-4.07550e+03
Coeff. C	-6.21220e+01
Temperature range (K), min.	329.28
Temperature range (K), max.	460.48

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C108496&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Critical temperatures and pressures of caprolactam, dimethyl sulfoxide, 1,4-dimethylpiperazine, and 2,6-dimethylpiperazine:	https://www.doi.org/10.1016/j.fluid.2018.05.029
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume

pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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