

# Benzaldehyde, 2,6-dichloro-

<b>Other names:</b>	2,6-Dichlorobenzaldehyde
<b>Inchi:</b>	InChI=1S/C7H4Cl2O/c8-6-2-1-3-7(9)5(6)4-10/h1-4H
<b>InchiKey:</b>	DMIYKWPEFRFTPY-UHFFFAOYSA-N
<b>Formula:</b>	C7H4Cl2O
<b>SMILES:</b>	O=Cc1c(Cl)cccc1Cl
<b>Mol. weight [g/mol]:</b>	175.01
<b>CAS:</b>	83-38-5

## Physical Properties

Property code	Value	Unit	Source
gf	-22.17	kJ/mol	Joback Method
hf	-91.28	kJ/mol	Joback Method
hfus	17.83	kJ/mol	Joback Method
hvap	50.27	kJ/mol	Joback Method
log10ws	-3.08		Crippen Method
logp	2.806		Crippen Method
mcvol	111.780	ml/mol	McGowan Method
pc	3960.52	kPa	Joback Method
tb	519.72	K	Joback Method
tc	753.22	K	Joback Method
tf	343.00 ± 3.00	K	NIST Webbook
vc	0.434	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	198.12	J/molxK	519.72	Joback Method
cpg	206.07	J/molxK	558.64	Joback Method
cpg	213.46	J/molxK	597.55	Joback Method
cpg	220.33	J/molxK	636.47	Joback Method
cpg	226.69	J/molxK	675.39	Joback Method
cpg	232.58	J/molxK	714.30	Joback Method
cpg	238.00	J/molxK	753.22	Joback Method
dvisc	0.0018573	Paxs	321.95	Joback Method

dvisc	0.0012274	Paxs	354.91	Joback Method
dvisc	0.0008704	Paxs	387.87	Joback Method
dvisc	0.0006513	Paxs	420.84	Joback Method
dvisc	0.0005084	Paxs	453.80	Joback Method
dvisc	0.0004103	Paxs	486.76	Joback Method
dvisc	0.0003403	Paxs	519.72	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C83385&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C83385&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/33-615-2/Benzaldehyde-2-6-dichloro.pdf>

Generated by Cheméo on 2024-04-25 18:58:02.310873091 +0000 UTC m=+16360731.231450402.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.