

# Cyclobuta[1,2-a:3,4-a']diacenaphthylene, 6b,6c,12b,12c-tetrahydro-

InChI: C1=CC2=C(C1)C=CC3=C2C=CC=C3  
InChIKey: TVMWOWUBPJYVIC-WAPCQROESA-N

Formula: C<sub>24</sub>H<sub>16</sub>  
SMILES: c1cc2c3c(cccc3c1)C1C2C2c3cccc4cccc(c34)C12  
Mol. weight [g/mol]: 304.38  
CAS: 15065-28-8

## Physical Properties

Property code	Value	Unit	Source
gf	785.84	kJ/mol	Joback Method
hf	499.49	kJ/mol	Joback Method
hfus	44.46	kJ/mol	Joback Method
hvap	78.07	kJ/mol	Joback Method
ie	7.58	eV	NIST Webbook
ie	7.58	eV	NIST Webbook
log10ws	-7.81		Crippen Method
logp	6.068		Crippen Method
mcvol	230.000	ml/mol	McGowan Method
pc	2069.88	kPa	Joback Method
tb	853.96	K	Joback Method
tc	1111.91	K	Joback Method
tf	595.74	K	Joback Method
vc	0.917	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	707.35	J/molxK	853.96	Joback Method
cpg	723.79	J/molxK	896.95	Joback Method
cpg	740.18	J/molxK	939.94	Joback Method
cpg	756.90	J/molxK	982.94	Joback Method
cpg	774.33	J/molxK	1025.93	Joback Method
cpg	792.87	J/molxK	1068.92	Joback Method
cpg	812.89	J/molxK	1111.91	Joback Method

dvisc	0.0358504	Paxs	595.74	Joback Method
dvisc	0.0402571	Paxs	638.78	Joback Method
dvisc	0.0445486	Paxs	681.81	Joback Method
dvisc	0.0487082	Paxs	724.85	Joback Method
dvisc	0.0527260	Paxs	767.89	Joback Method
dvisc	0.0565971	Paxs	810.92	Joback Method
dvisc	0.0603200	Paxs	853.96	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15065288&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15065288&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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