

# Acetamide, N-(4-formylphenyl)-

|                             |   |
|-----------------------------|---|
| <b>Other names:</b>         | 4'-Formylacetanilide<br>4-(Acetalamino)benzaldehyde<br>4-Acetamidobenzaldehyde<br>4-Acetomidobenzaldehyde<br>4-Acetylaminobenzaldehyde<br>4-Formylacetanilide<br>Acetanilide, 4'-formyl-<br>Benzaldehyde, 4-acetamido-<br>Micotiazone<br>NSC 1701<br>p-(Acetylamino)benzaldehyde<br>p-Acetamidobenzaldehyde<br>p-Acetaminobenzaldehyde<br>p-Formylacetanilide<br>para-Acetaminobenzaldehyde |
| <b>Inchi:</b>               | InChI=1S/C9H9NO2/c1-7(12)10-9-4-2-8(6-11)3-5-9/h2-6H,1H3,(H,10,12)  |
| <b>InchiKey:</b>            | SKLUWKYNZNXLX-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C9H9NO2   |
| <b>SMILES:</b>              | CC(=O)Nc1ccc(C=O)cc1  |
| <b>Mol. weight [g/mol]:</b> | 163.17  |
| <b>CAS:</b>                 | 122-85-0  |

## Physical Properties

| Property code | Value   | Unit   | Source                               |
|---------------|---------|--------|--------------------------------------|
| gf            | -11.37  | kJ/mol | Joback Method                        |
| hf            | -148.72 | kJ/mol | Joback Method                        |
| hfus          | 21.71   | kJ/mol | Joback Method                        |
| hvap          | 58.47   | kJ/mol | Joback Method                        |
| log10ws       | -1.58   |        | Aqueous Solubility Prediction Method |
| logp          | 1.458   |        | Crippen Method                       |
| mcvol         | 127.030 | ml/mol | McGowan Method                       |
| pc            | 3829.28 | kPa    | Joback Method                        |
| tb            | 589.68  | K      | Joback Method                        |
| tc            | 812.41  | K      | Joback Method                        |
| tf            | 430.82  | K      | Aqueous Solubility Prediction Method |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 293.04 | J/mol×K | 589.68          | Joback Method |
| cpg           | 304.13 | J/mol×K | 626.80          | Joback Method |
| cpg           | 314.46 | J/mol×K | 663.92          | Joback Method |
| cpg           | 324.07 | J/mol×K | 701.04          | Joback Method |
| cpg           | 332.99 | J/mol×K | 738.16          | Joback Method |
| cpg           | 341.25 | J/mol×K | 775.29          | Joback Method |
| cpg           | 348.88 | J/mol×K | 812.41          | Joback Method |
| hsubt         | 99.00  | kJ/mol  | 337.00          | NIST Webbook  |

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C122850&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hsubt:</b>   | Enthalpy of sublimation at a given temperature  |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mccvol:</b>  | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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