

# Benzene, 4-methoxy-1-methyl-2-nitro-

<b>Other names:</b>	Anisole, 4-methyl-3-nitro- 4-Methoxy-2-nitrotoluene 4-Methyl-3-nitroanisole 2-Nitro-4-methoxytoluene
<b>Inchi:</b>	InChI=1S/C8H9NO3/c1-6-3-4-7(12-2)5-8(6)9(10)11/h3-5H,1-2H3
<b>InchiKey:</b>	JBORNNGTJSTLC-UHFFFAOYSA-N
<b>Formula:</b>	C8H9NO3
<b>SMILES:</b>	COc1ccc(C)c([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	167.16
<b>CAS:</b>	17484-36-5

## Physical Properties

Property code	Value	Unit	Source
gf	40.18	kJ/mol	Joback Method
hf	-137.84	kJ/mol	Joback Method
hfus	22.29	kJ/mol	Joback Method
hvap	56.00	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	1.912		Crippen Method
mcvol	123.110	ml/mol	McGowan Method
pc	3501.28	kPa	Joback Method
tb	539.70	K	NIST Webbook
tc	835.30	K	Joback Method
tf	397.22	K	Joback Method
vc	0.475	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.57	J/molxK	593.34	Joback Method
cpg	298.32	J/molxK	633.67	Joback Method
cpg	309.33	J/molxK	673.99	Joback Method
cpg	319.61	J/molxK	714.32	Joback Method
cpg	329.15	J/molxK	754.64	Joback Method

cpg	337.99	J/mol×K	794.97	Joback Method
cpg	346.11	J/mol×K	835.30	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17484365&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17484365&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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