

# Benzene, (bromomethyl)-

<b>Other names:</b>	(Bromomethyl)benzene 1-Bromotoluene Benzyl bromide Bromophenylmethane Bromotoluene, «alpha» Bromotoluene, Â«alphaÂ» NSC 8041 Phenylmethyl bromide Toluene, «alpha»-bromo- Toluene, Â«alphaÂ»-bromo- UN 1737 «alpha»-Bromotoluene «omega»-Bromotoluene Â«alphaÂ»-Bromotoluene Â«omegaÂ»-Bromotoluene
<b>Inchi:</b>	InChI=1S/C7H7Br/c8-6-7-4-2-1-3-5-7/h1-5H,6H2
<b>InchiKey:</b>	AGEZXYOZHKGVCN-UHFFFAOYSA-N
<b>Formula:</b>	C7H7Br
<b>SMILES:</b>	BrCc1ccccc1
<b>Mol. weight [g/mol]:</b>	171.03
<b>CAS:</b>	100-39-0

## Physical Properties

Property code	Value	Unit	Source
gf	134.79	kJ/mol	Joback Method
hf	84.00 ± 4.00	kJ/mol	NIST Webbook
hfl	16.00 ± 2.00	kJ/mol	NIST Webbook
hfl	22.00	kJ/mol	NIST Webbook
hfl	30.00 ± 14.00	kJ/mol	NIST Webbook
hfus	13.21	kJ/mol	Joback Method
hvap	53.70	kJ/mol	NIST Webbook
hvap	50.50 ± 0.50	kJ/mol	NIST Webbook
hvap	47.30 ± 4.20	kJ/mol	NIST Webbook
hvap	53.30 ± 0.70	kJ/mol	NIST Webbook
ie	9.10 ± 0.10	eV	NIST Webbook
ie	9.23	eV	NIST Webbook
ie	9.02 ± 0.02	eV	NIST Webbook

ie	8.99 ± 0.01	eV	NIST Webbook
ie	9.10 ± 0.05	eV	NIST Webbook
ie	8.90	eV	NIST Webbook
ie	8.99 ± 0.01	eV	NIST Webbook
log10ws	-2.78		Crippen Method
logp	2.582		Crippen Method
mcvol	103.230	ml/mol	McGowan Method
pc	4504.30	kPa	Joback Method
rinpol	1087.00		NIST Webbook
rinpol	1087.00		NIST Webbook
rinpol	1059.00		NIST Webbook
rinpol	1061.00		NIST Webbook
tb	471.00	K	NIST Webbook
tb	471.00	K	NIST Webbook
tb	471.70	K	NIST Webbook
tc	684.64	K	Joback Method
tf	271.75 ± 0.50	K	NIST Webbook
vc	0.382	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.09	J/molxK	452.40	Joback Method
cpg	184.03	J/molxK	491.11	Joback Method
cpg	194.16	J/molxK	529.81	Joback Method
cpg	203.55	J/molxK	568.52	Joback Method
cpg	212.23	J/molxK	607.23	Joback Method
cpg	220.26	J/molxK	645.94	Joback Method
cpg	227.67	J/molxK	684.64	Joback Method
dvisc	0.0010283	Paxs	320.71	Joback Method
dvisc	0.0016279	Paxs	287.79	Joback Method
dvisc	0.0029016	Paxs	254.87	Joback Method
dvisc	0.0007076	Paxs	353.63	Joback Method
dvisc	0.0005189	Paxs	386.56	Joback Method
dvisc	0.0003995	Paxs	419.48	Joback Method
dvisc	0.0003195	Paxs	452.40	Joback Method
hfust	13.20	kJ/mol	271.80	NIST Webbook
hfust	13.20	kJ/mol	271.80	NIST Webbook
hvapt	46.90	kJ/mol	388.50	NIST Webbook
hvapt	48.10	kJ/mol	374.50	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	400.00	K	11.00	NIST Webbook
tbrp	400.00	K	11.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42835e+01
Coeff. B	-3.85387e+03
Coeff. C	-7.29640e+01
Temperature range (K), min.	348.32
Temperature range (K), max.	502.50

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C100390&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C100390&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvsc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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