

# Acetamide, N-tetrahydrofurfuryl-2,2-dichloro-

<b>Inchi:</b>	InChI=1S/C7H11Cl2NO2/c8-6(9)7(11)10-4-5-2-1-3-12-5/h5-6H,1-4H2,(H,10,11)
<b>InchiKey:</b>	JUNUKZRYFJPFGM-UHFFFAOYSA-N
<b>Formula:</b>	C7H11Cl2NO2
<b>SMILES:</b>	O=C(NCC1CCCO1)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	212.07

## Physical Properties

Property code	Value	Unit	Source
gf	-107.34	kJ/mol	Joback Method
hf	-355.20	kJ/mol	Joback Method
hfus	27.37	kJ/mol	Joback Method
hvap	57.51	kJ/mol	Joback Method
log10ws	-1.72		Crippen Method
logp	1.085		Crippen Method
mcvol	140.530	ml/mol	McGowan Method
pc	3423.86	kPa	Joback Method
rinpola	1468.00		NIST Webbook
tb	580.25	K	Joback Method
tc	802.85	K	Joback Method
tf	353.55	K	Joback Method
vc	0.522	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.16	J/mol×K	580.25	Joback Method
cpg	338.93	J/mol×K	617.35	Joback Method
cpg	350.83	J/mol×K	654.45	Joback Method
cpg	361.89	J/mol×K	691.55	Joback Method
cpg	372.14	J/mol×K	728.65	Joback Method
cpg	381.63	J/mol×K	765.75	Joback Method
cpg	390.39	J/mol×K	802.85	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307296&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307296&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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