

Valerenyl acetate

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C17H26O2/c1-11(10-19-14(4)18)9-15-7-5-12(2)16-8-6-13(3)17(15)16/h9,12,15

GMTBKCJRMNJNFL-HEEVJLJPSA-N

C17H26O2

CC(=O)OCC(C)=CC1CCC(C)C2CCC(C)=C12

262.39

Physical Properties

Property code	Value	Unit	Source
gf	18.20	kJ/mol	Joback Method
hf	-389.96	kJ/mol	Joback Method
hfus	32.95	kJ/mol	Joback Method
hvap	64.28	kJ/mol	Joback Method
log10ws	-4.57		Crippen Method
logp	4.268		Crippen Method
mcvol	227.510	ml/mol	McGowan Method
pc	1681.03	kPa	Joback Method
rinpol	1897.00		NIST Webbook
rinpol	1897.00		NIST Webbook
ripol	2290.00		NIST Webbook
tb	699.43	K	Joback Method
tc	911.60	K	Joback Method
tf	381.35	K	Joback Method
vc	0.868	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	669.84	J/mol×K	699.43	Joback Method
cpg	690.13	J/mol×K	734.79	Joback Method
cpg	709.20	J/mol×K	770.15	Joback Method
cpg	727.09	J/mol×K	805.51	Joback Method
cpg	743.88	J/mol×K	840.88	Joback Method
cpg	759.60	J/mol×K	876.24	Joback Method
cpg	774.33	J/mol×K	911.60	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R203913&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
ripola:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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