

# 5,7-dibromo-2-(5,7-dibromo-1,3-dihydro-3-oxo-2H-indol-2-ylidene)-1,2-dihydro-3H-indol-1-one

## Other names:

Inchi:	5,7-Dibromo-2-(5,7-dibromo-1,3-dihydro-3-oxo-2H-indol-2-ylidene)-1,2-dihydro-3H-indol-1-one (C.I. Vat Blue 5)
InchiKey:	PTWYQANXSNMUTI-BUHFOSPRSA-N
Formula:	C16H6Br4N2O2
SMILES:	O=C1C(=C2Nc3c(Br)cc(Br)cc3C2=O)Nc2c(Br)cc(Br)cc21
Mol. weight [g/mol]:	577.85
CAS:	2475-31-2

## Physical Properties

Property code	Value	Unit	Source
gf	386.02	kJ/mol	Joback Method
hf	157.33	kJ/mol	Joback Method
hfus	56.86	kJ/mol	Joback Method
hvap	109.54	kJ/mol	Joback Method
log10ws	-8.91		Crippen Method
logp	5.865		Crippen Method
mcvol	255.860	ml/mol	McGowan Method
pc	4730.11	kPa	Joback Method
tb	1178.04	K	Joback Method
tc	1495.61	K	Joback Method
tf	1053.90	K	Joback Method
vc	0.954	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	662.90	J/molxK	1442.68	Joback Method
cpg	627.14	J/molxK	1178.04	Joback Method
cpg	635.06	J/molxK	1230.97	Joback Method
cpg	642.47	J/molxK	1283.90	Joback Method
cpg	649.51	J/molxK	1336.82	Joback Method
cpg	656.27	J/molxK	1389.75	Joback Method
cpg	669.50	J/molxK	1495.61	Joback Method
hsubt	129.00	kJ/mol	576.50	NIST Webbook

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2475312&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2475312&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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