

# Fumaric acid, 2-decyl heptadecyl ester

<b>Inchi:</b>	InChI=1S/C31H58O4/c1-4-6-8-10-12-13-14-15-16-17-18-19-20-22-24-28-34-30(32)26-27
<b>InchiKey:</b>	WRRIPMQXRPCYPL-CYYJNZCTSA-N
<b>Formula:</b>	C31H58O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OC(C)CCCCCCCC
<b>Mol. weight [g/mol]:</b>	494.79

## Physical Properties

Property code	Value	Unit	Source
gf	-179.92	kJ/mol	Joback Method
hf	-1060.83	kJ/mol	Joback Method
hfus	78.30	kJ/mol	Joback Method
hvap	102.48	kJ/mol	Joback Method
log10ws	-10.49		Crippen Method
logp	9.639		Crippen Method
mvol	458.230	ml/mol	McGowan Method
pc	612.08	kPa	Joback Method
rinpol	3386.00		NIST Webbook
rinpol	3386.00		NIST Webbook
tb	1064.98	K	Joback Method
tc	1338.92	K	Joback Method
tf	563.37	K	Joback Method
vc	1.794	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1642.70	J/molxK	1064.98	Joback Method
cpg	1667.05	J/molxK	1110.64	Joback Method
cpg	1689.04	J/molxK	1156.29	Joback Method
cpg	1708.83	J/molxK	1201.95	Joback Method
cpg	1726.56	J/molxK	1247.61	Joback Method
cpg	1742.40	J/molxK	1293.26	Joback Method
cpg	1756.51	J/molxK	1338.92	Joback Method
dvisc	0.0002169	Paxs	563.37	Joback Method

dvisc	0.0000870	Paxs	646.97	Joback Method
dvisc	0.0000431	Paxs	730.57	Joback Method
dvisc	0.0000246	Paxs	814.17	Joback Method
dvisc	0.0000156	Paxs	897.78	Joback Method
dvisc	0.0000107	Paxs	981.38	Joback Method
dvisc	0.0000078	Paxs	1064.98	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348598&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348598&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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