

# Butanoic acid, 2-chloro, 1-methylethyl ester

**Inchi:** InChI=1S/C7H13ClO2/c1-4-6(8)7(9)10-5(2)3/h5-6H,4H2,1-3H3  
**InchiKey:** DAOWVENIDWCKPM-UHFFFAOYSA-N  
**Formula:** C7H13ClO2  
**SMILES:** CCC(Cl)C(=O)OC(C)C  
**Mol. weight [g/mol]:** 164.63

## Physical Properties

Property code	Value	Unit	Source
gf	-242.67	kJ/mol	Joback Method
hf	-458.91	kJ/mol	Joback Method
hfus	13.82	kJ/mol	Joback Method
hvap	43.94	kJ/mol	Joback Method
log10ws	-1.99		Crippen Method
logp	1.955		Crippen Method
mcvol	129.170	ml/mol	McGowan Method
pc	2875.03	kPa	Joback Method
ripol	967.00		NIST Webbook
ripol	952.00		NIST Webbook
ripol	973.00		NIST Webbook
ripol	977.00		NIST Webbook
ripol	959.00		NIST Webbook
ripol	1298.00		NIST Webbook
ripol	1291.00		NIST Webbook
ripol	1304.00		NIST Webbook
ripol	1304.00		NIST Webbook
ripol	1327.00		NIST Webbook
tb	472.40	K	Joback Method
tc	663.16	K	Joback Method
tf	240.73	K	Joback Method
vc	0.488	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	265.59	J/molxK	472.40	Joback Method
cpg	277.03	J/molxK	504.19	Joback Method
cpg	288.00	J/molxK	535.99	Joback Method
cpg	298.52	J/molxK	567.78	Joback Method
cpg	308.57	J/molxK	599.57	Joback Method
cpg	318.17	J/molxK	631.37	Joback Method
cpg	327.32	J/molxK	663.16	Joback Method
dvisc	0.0064287	Paxs	240.73	Joback Method
dvisc	0.0025878	Paxs	279.34	Joback Method
dvisc	0.0012993	Paxs	317.95	Joback Method
dvisc	0.0007574	Paxs	356.56	Joback Method
dvisc	0.0004906	Paxs	395.18	Joback Method
dvisc	0.0003433	Paxs	433.79	Joback Method
dvisc	0.0002547	Paxs	472.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R28590&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R28590&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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