

Nonane, 2,2,3-trimethyl-

Other names:	2,2,3-Trimethylnonane
Inchi:	InChI=1S/C12H26/c1-6-7-8-9-10-11(2)12(3,4)5/h11H,6-10H2,1-5H3
InchiKey:	XXNUJUNKYOZLAJ-UHFFFAOYSA-N
Formula:	C12H26
SMILES:	CCCCCCC(C)C(C)(C)C
Mol. weight [g/mol]:	170.33
CAS:	55499-04-2

Physical Properties

Property code	Value	Unit	Source
gf	50.56	kJ/mol	Joback Method
hf	-305.04	kJ/mol	Joback Method
hfus	15.90	kJ/mol	Joback Method
hvap	40.62	kJ/mol	Joback Method
log10ws	-4.36		Crippen Method
logp	4.639		Crippen Method
mcvol	179.940	ml/mol	McGowan Method
pc	1821.61	kPa	Joback Method
rinpol	1114.00		NIST Webbook
tb	470.29	K	Joback Method
tc	644.11	K	Joback Method
tf	212.42	K	Joback Method
vc	0.691	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	409.02	J/molxK	470.29	Joback Method
cpg	427.54	J/molxK	499.26	Joback Method
cpg	445.22	J/molxK	528.23	Joback Method
cpg	462.09	J/molxK	557.20	Joback Method
cpg	478.18	J/molxK	586.17	Joback Method
cpg	493.52	J/molxK	615.14	Joback Method
cpg	508.14	J/molxK	644.11	Joback Method

dvisc	0.0170934	Paxs	212.42	Joback Method
dvisc	0.0043803	Paxs	255.40	Joback Method
dvisc	0.0016616	Paxs	298.38	Joback Method
dvisc	0.0008046	Paxs	341.36	Joback Method
dvisc	0.0004582	Paxs	384.33	Joback Method
dvisc	0.0002922	Paxs	427.31	Joback Method
dvisc	0.0002023	Paxs	470.29	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38156e+01
Coeff. B	-3.51185e+03
Coeff. C	-9.13150e+01
Temperature range (K), min.	350.91
Temperature range (K), max.	504.27

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C55499042&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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