

# 4,5-Octanedione

<b>Other names:</b>	4,5-Octadione Bibutyryl n-Octane-4,5-dione octane-4,5-dione
<b>Inchi:</b>	InChI=1S/C8H14O2/c1-3-5-7(9)8(10)6-4-2/h3-6H2,1-2H3
<b>InchiKey:</b>	XYZAPOXYXNIBEU-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O2
<b>SMILES:</b>	CCCC(=O)C(=O)CCC
<b>Mol. weight [g/mol]:</b>	142.20
<b>CAS:</b>	5455-24-3

## Physical Properties

Property code	Value	Unit	Source
gf	-241.36	kJ/mol	Joback Method
hf	-433.61	kJ/mol	Joback Method
hfus	19.67	kJ/mol	Joback Method
hvap	46.89	kJ/mol	Joback Method
log10ws	-1.73		Crippen Method
logp	1.725		Crippen Method
mcvol	126.720	ml/mol	McGowan Method
pc	2890.51	kPa	Joback Method
tb	441.20	K	NIST Webbook
tc	675.94	K	Joback Method
tf	279.78	K	Joback Method
vc	0.495	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.20	J/molxK	490.18	Joback Method
cpg	286.97	J/molxK	521.14	Joback Method
cpg	298.23	J/molxK	552.10	Joback Method
cpg	308.97	J/molxK	583.06	Joback Method
cpg	319.22	J/molxK	614.02	Joback Method

cpg	328.98	J/molxK	644.98	Joback Method
cpg	338.27	J/molxK	675.94	Joback Method
dvisc	0.0036359	Paxs	279.78	Joback Method
dvisc	0.0019569	Paxs	314.85	Joback Method
dvisc	0.0011925	Paxs	349.91	Joback Method
dvisc	0.0007953	Paxs	384.98	Joback Method
dvisc	0.0005675	Paxs	420.05	Joback Method
dvisc	0.0004266	Paxs	455.11	Joback Method
dvisc	0.0003340	Paxs	490.18	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	333.20	K	1.60	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49754e+01
Coeff. B	-3.90888e+03
Coeff. C	-6.37900e+01
Temperature range (K), min.	329.92
Temperature range (K), max.	468.27

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C5455243&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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