

# 1,2-Diiodotetrafluoroethane

<b>Other names:</b>	1,1,2,2-tetrafluoro-1,2-diiodoethane
<b>Inchi:</b>	InChI=1S/C2F4I2/c3-1(4,7)2(5,6)8
<b>InchiKey:</b>	NZXVPCQHQVWOFD-UHFFFAOYSA-N
<b>Formula:</b>	C2F4I2
<b>SMILES:</b>	FC(F)(I)C(F)(F)I
<b>Mol. weight [g/mol]:</b>	353.82
<b>CAS:</b>	354-65-4

## Physical Properties

Property code	Value	Unit	Source
gf	-691.36	kJ/mol	Joback Method
hf	-667.00 ± 3.00	kJ/mol	NIST Webbook
hfus	7.24	kJ/mol	Joback Method
hvap	32.93	kJ/mol	Joback Method
ie	10.11 ± 0.01	eV	NIST Webbook
log10ws	-4.17		Crippen Method
logp	3.042		Crippen Method
mcvol	97.760	ml/mol	McGowan Method
pc	3925.85	kPa	Joback Method
rinpola	790.00		NIST Webbook
rinpola	794.00		NIST Webbook
rinpola	791.00		NIST Webbook
tb	386.00 ± 2.00	K	NIST Webbook
tb	385.50 ± 0.50	K	NIST Webbook
tc	651.53	K	Joback Method
tf	235.62	K	Joback Method
vc	0.373	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	143.11	J/mol×K	422.06	Joback Method
cpg	148.54	J/mol×K	460.30	Joback Method
cpg	153.09	J/mol×K	498.55	Joback Method

cpg	156.85	J/mol×K	536.79	Joback Method
cpg	159.91	J/mol×K	575.04	Joback Method
cpg	162.35	J/mol×K	613.28	Joback Method
cpg	164.26	J/mol×K	651.53	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C354654&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C354654&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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