

# «alpha»-Naphthyl red

<b>Other names:</b>	1-Naphthalenamine, 4-(phenylazo)- C.I. Solvent Yellow 4 C.I. 11350 Naphthalene, 1-phenylazo-4-amino- Nubian Yellow TB Phenylazo-«alpha»-naphthylamine 1-Naphthylamine, 4-(phenylazo)- 4-(Phenylazo)-1-naphthylamine 4-Benzene-azo-«alpha»-naphthylamine 4-Benzeneazo-1-naphthylamine 1-Amino-4-phenylazonaphthalene 1-Naphthalenamine, 4-(2-phenyldiazenyl)- NSC 13974
<b>Inchi:</b>	InChI=1S/C16H13N3/c17-15-10-11-16(14-9-5-4-8-13(14)15)19-18-12-6-2-1-3-7-12/h1-11
<b>InchiKey:</b>	IICHURGZQPGTRD-UHFFFAOYSA-N
<b>Formula:</b>	C16H13N3
<b>SMILES:</b>	<chem>Nc1ccc(N=Nc2ccccc2)c2ccccc12</chem>
<b>Mol. weight [g/mol]:</b>	247.29
<b>CAS:</b>	131-22-6

## Physical Properties

Property code	Value	Unit	Source
hf	348.63	kJ/mol	Joback Method
hvap	76.04	kJ/mol	Joback Method
log10ws	-4.95		Crippen Method
logp	4.837		Crippen Method
mvol	194.960	ml/mol	McGowan Method
pc	2306.95	kPa	Joback Method
tb	869.51	K	Joback Method
tc	1149.33	K	Joback Method

## Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C131226&Units=SI>

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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