

# 2-Ethylbutyric acid, 2-isopropylphenyl ester

<b>Inchi:</b>	InChI=1S/C15H22O2/c1-5-12(6-2)15(16)17-14-10-8-7-9-13(14)11(3)4/h7-12H,5-6H2,1-4
<b>InchiKey:</b>	CQTLCBJYJHAEBR-UHFFFAOYSA-N
<b>Formula:</b>	C15H22O2
<b>SMILES:</b>	CCC(CC)C(=O)Oc1ccccc1C(C)C
<b>Mol. weight [g/mol]:</b>	234.33

## Physical Properties

Property code	Value	Unit	Source
gf	-60.60	kJ/mol	Joback Method
hf	-383.23	kJ/mol	Joback Method
hfus	24.00	kJ/mol	Joback Method
hvap	60.30	kJ/mol	Joback Method
log10ws	-4.40		Crippen Method
logp	4.152		Crippen Method
mcvol	205.890	ml/mol	McGowan Method
pc	1923.67	kPa	Joback Method
rinsol	1532.00		NIST Webbook
tb	649.67	K	Joback Method
tc	855.34	K	Joback Method
tf	339.91	K	Joback Method
vc	0.779	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	547.01	J/molxK	649.67	Joback Method
cpg	564.30	J/molxK	683.95	Joback Method
cpg	580.59	J/molxK	718.23	Joback Method
cpg	595.92	J/molxK	752.51	Joback Method
cpg	610.31	J/molxK	786.79	Joback Method
cpg	623.79	J/molxK	821.07	Joback Method
cpg	636.37	J/molxK	855.34	Joback Method
dvisc	0.0023739	Paxs	339.91	Joback Method
dvisc	0.0010314	Paxs	391.54	Joback Method

dvisc	0.0005442	Paxs	443.16	Joback Method
dvisc	0.0003281	Paxs	494.79	Joback Method
dvisc	0.0002177	Paxs	546.42	Joback Method
dvisc	0.0001550	Paxs	598.04	Joback Method
dvisc	0.0001165	Paxs	649.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369986&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369986&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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