

# 2-(isopropyl(trifluoroacetyl)amino)ethyl trifluoroacetate

**Other names:** 2-(N-Isopropyl-trifluoroacetyl-amino)ethanol trifluoroacetate

(N-Trifluoroacetyloxyethyl)isopropyl-trifluoroacetylamine

**Inchi:** InChI=1S/C9H11F6NO3/c1-5(2)16(6(17)8(10,11)12)3-4-19-7(18)9(13,14)15/h5H,3-4H2,1

**InchiKey:** KTOOQRYBIWTNRO-UHFFFAOYSA-N

**Formula:** C9H11F6NO3

**SMILES:** CC(C)N(CCOC(=O)C(F)(F)F)C(=O)C(F)(F)F

**Mol. weight [g/mol]:** 295.18

## Physical Properties

Property code	Value	Unit	Source
gf	-1392.78	kJ/mol	Joback Method
hf	-1718.38	kJ/mol	Joback Method
hfus	26.60	kJ/mol	Joback Method
hvap	45.69	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	1.891		Crippen Method
mcvol	167.280	ml/mol	McGowan Method
pc	2068.00	kPa	Joback Method
rinpol	1067.00		NIST Webbook
tb	536.64	K	Joback Method
tc	694.65	K	Joback Method
tf	339.13	K	Joback Method
vc	0.667	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	434.37	J/molxK	536.64	Joback Method
cpg	446.09	J/molxK	562.97	Joback Method
cpg	457.14	J/molxK	589.31	Joback Method
cpg	467.56	J/molxK	615.64	Joback Method
cpg	477.38	J/molxK	641.98	Joback Method
cpg	486.61	J/molxK	668.31	Joback Method
cpg	495.29	J/molxK	694.65	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373297&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373297&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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