

# Hexadecanoic acid, 2-hydroxy-6-methyl, methyl ester

Inchi:	InChI=1S/C18H36O3/c1-4-5-6-7-8-9-10-11-13-16(2)14-12-15-17(19)18(20)21-3/h16-17,1
InchiKey:	OLJWYDQEQRNRTLL-UHFFFAOYSA-N
Formula:	C18H36O3
SMILES:	CCCCCCCCC(C)CCCC(O)C(=O)OC
Mol. weight [g/mol]:	300.48

## Physical Properties

Property code	Value	Unit	Source
gf	-274.94	kJ/mol	Joback Method
hf	-822.44	kJ/mol	Joback Method
hfus	42.20	kJ/mol	Joback Method
hvap	80.72	kJ/mol	Joback Method
log10ws	-5.35		Crippen Method
logp	4.858		Crippen Method
mcvol	277.790	ml/mol	McGowan Method
pc	1282.83	kPa	Joback Method
rinsol	2067.00		NIST Webbook
tb	778.83	K	Joback Method
tc	956.64	K	Joback Method
tf	395.60	K	Joback Method
vc	1.075	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	867.49	J/molxK	778.83	Joback Method
cpg	944.67	J/molxK	927.00	Joback Method
cpg	930.91	J/molxK	897.37	Joback Method
cpg	916.33	J/molxK	867.73	Joback Method
cpg	900.92	J/molxK	838.10	Joback Method
cpg	884.65	J/molxK	808.46	Joback Method
cpg	957.64	J/molxK	956.64	Joback Method
dvisc	0.0000150	Paxs	778.83	Joback Method
dvisc	0.0000242	Paxs	714.96	Joback Method

dvisc	0.0000429	Paxs	651.09	Joback Method
dvisc	0.0000860	Paxs	587.22	Joback Method
dvisc	0.0002045	Paxs	523.34	Joback Method
dvisc	0.0006189	Paxs	459.47	Joback Method
dvisc	0.0026782	Paxs	395.60	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R98145&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R98145&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/34-304-6/Hexadecanoic-acid-2-hydroxy-6-methyl-methyl-ester.pdf>

Generated by Cheméo on 2024-04-30 14:00:04.696409548 +0000 UTC m=+16774853.616986859.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.