

# Resorcinol monoacetate

<b>Other names:</b>	1,3-Benzenediol, monoacetate m-Hydroxyphenyl acetate Acetylresorcinol Euresol Remonol Resorcin monoacetate Resorcitate m-Acetoxyphenol 3-Acetoxyphenol Resorcinol, acetate Acetic acid, 3-hydroxyphenyl ester 1,3-Benzenediol, 1-acetate NSC 40511 1,3-Benzendiol monoacetate 3-Hydroxyphenyl acetate
<b>Inchi:</b>	InChI=1S/C8H8O3/c1-6(9)11-8-4-2-3-7(10)5-8/h2-5,10H,1H3
<b>InchiKey:</b>	ZZPKZRHERLGEKA-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O3
<b>SMILES:</b>	CC(=O)Oc1cccc(O)c1
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	102-29-4

## Physical Properties

Property code	Value	Unit	Source
gf	-259.65	kJ/mol	Joback Method
hf	-394.03	kJ/mol	Joback Method
hfus	19.09	kJ/mol	Joback Method
hvap	57.85	kJ/mol	Joback Method
log10ws	-1.33		Crippen Method
logp	1.317		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	4691.31	kPa	Joback Method
rinpol	1382.60		NIST Webbook
tb	556.00	K	NIST Webbook
tb	556.20	K	NIST Webbook
tc	798.25	K	Joback Method
tf	390.22	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.25	J/molxK	566.03	Joback Method
cpg	269.54	J/molxK	604.73	Joback Method
cpg	279.09	J/molxK	643.44	Joback Method
cpg	287.96	J/molxK	682.14	Joback Method
cpg	296.23	J/molxK	720.84	Joback Method
cpg	303.95	J/molxK	759.54	Joback Method
cpg	311.19	J/molxK	798.25	Joback Method
dvisc	0.0011881	Paxs	390.22	Joback Method
dvisc	0.0005829	Paxs	419.52	Joback Method
dvisc	0.0003139	Paxs	448.82	Joback Method
dvisc	0.0001823	Paxs	478.12	Joback Method
dvisc	0.0001128	Paxs	507.43	Joback Method
dvisc	0.0000735	Paxs	536.73	Joback Method
dvisc	0.0000501	Paxs	566.03	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C102294&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C102294&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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