

# Lanceol oxide

<b>Inchi:</b>	InChI=1S/C14H22O/c1-11-6-8-13(9-7-11)14-5-3-4-12(2)10-15-14/h4,6,13-14H,3,5,7-10H
<b>InchiKey:</b>	QTWUDCCAQVTTQQ-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O
<b>SMILES:</b>	CC1=CCC(C2CCC=C(C)CO2)CC1
<b>Mol. weight [g/mol]:</b>	206.32

## Physical Properties

Property code	Value	Unit	Source
gf	58.34	kJ/mol	Joback Method
hf	-269.19	kJ/mol	Joback Method
hfus	23.23	kJ/mol	Joback Method
hvap	54.21	kJ/mol	Joback Method
log10ws	-4.14		Crippen Method
logp	3.858		Crippen Method
mvol	183.670	ml/mol	McGowan Method
pc	2280.59	kPa	Joback Method
rinpol	1695.00		NIST Webbook
rinpol	1695.00		NIST Webbook
tb	598.32	K	Joback Method
tc	833.87	K	Joback Method
tf	311.91	K	Joback Method
vc	0.670	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	490.57	J/molxK	598.32	Joback Method
cpg	514.22	J/molxK	637.58	Joback Method
cpg	536.34	J/molxK	676.84	Joback Method
cpg	556.95	J/molxK	716.10	Joback Method
cpg	576.09	J/molxK	755.36	Joback Method
cpg	593.79	J/molxK	794.62	Joback Method
cpg	610.10	J/molxK	833.87	Joback Method
dvisc	0.0035446	Paxs	311.91	Joback Method

dvisc	0.0014984	Paxs	359.64	Joback Method
dvisc	0.0007750	Paxs	407.38	Joback Method
dvisc	0.0004603	Paxs	455.11	Joback Method
dvisc	0.0003018	Paxs	502.85	Joback Method
dvisc	0.0002129	Paxs	550.58	Joback Method
dvisc	0.0001588	Paxs	598.32	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R407239&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R407239&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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