

Ethanone, 1-(2-hydroxy-1-naphthalenyl)-

Other names:	1'-Acetonaphthone, 2'-hydroxy-1-Acetyl-2-hydroxynaphthlene 1-Acetyl-2-naphthol 2'-Hydroxy-1'-acetonaphthone 1-Acetylnaphthalen-2-ol 1-(2-hydroxy-1-naphthyl)ethan-1-one
Inchi:	InChI=1S/C12H10O2/c1-8(13)12-10-5-3-2-4-9(10)6-7-11(12)14/h2-7,14H,1H3
InchiKey:	VUIOUIWZVKVFCI-UHFFFAOYSA-N
Formula:	C11H10O2
SMILES:	CC(=O)c1c(O)ccc2ccccc12
Mol. weight [g/mol]:	174.20
CAS:	574-19-6

Physical Properties

Property code	Value	Unit	Source
gf	-23.95	kJ/mol	Joback Method
hf	-164.77	kJ/mol	Joback Method
hfus	24.89	kJ/mol	Joback Method
hvap	66.64	kJ/mol	Joback Method
log10ws	-3.48		Crippen Method
logp	2.748		Crippen Method
mcvol	144.160	ml/mol	McGowan Method
pc	3920.94	kPa	Joback Method
tb	659.09	K	Joback Method
tc	906.83	K	Joback Method
tf	336.90 ± 0.01	K	NIST Webbook
vc	0.493	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	409.89	J/molxK	865.54	Joback Method
cpg	418.49	J/molxK	906.83	Joback Method
cpg	359.32	J/molxK	659.09	Joback Method

cpg	370.92	J/mol×K	700.38	Joback Method
cpg	381.63	J/mol×K	741.67	Joback Method
cpg	391.60	J/mol×K	782.96	Joback Method
cpg	400.97	J/mol×K	824.25	Joback Method
dvisc	0.0000441	Paxs	659.09	Joback Method
dvisc	0.0000604	Paxs	625.62	Joback Method
dvisc	0.0005830	Paxs	458.29	Joback Method
dvisc	0.0003275	Paxs	491.76	Joback Method
dvisc	0.0001980	Paxs	525.22	Joback Method
dvisc	0.0001271	Paxs	558.69	Joback Method
dvisc	0.0000858	Paxs	592.16	Joback Method
hfust	21.34	kJ/mol	337.00	NIST Webbook
hfust	21.34	kJ/mol	337.00	NIST Webbook

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C574196&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

vc: Critical Volume

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