

# Hexanoic acid, 2-bromo-

<b>Other names:</b>	«alpha»-Bromo-n-caproic acid «alpha»-Bromocaproic acid «alpha»-Bromohexanoic acid 2-Bromocaproic acid 2-Bromohexanoic acid
<b>Inchi:</b>	InChI=1S/C6H11BrO2/c1-2-3-4-5(7)6(8)9/h5H,2-4H2,1H3,(H,8,9)
<b>InchiKey:</b>	HZTPKMIMXLTOSK-UHFFFAOYSA-N
<b>Formula:</b>	C6H11BrO2
<b>SMILES:</b>	CCCCC(Br)C(=O)O
<b>Mol. weight [g/mol]:</b>	195.05
<b>CAS:</b>	616-05-7

## Physical Properties

Property code	Value	Unit	Source
gf	-254.22	kJ/mol	Joback Method
hf	-410.93	kJ/mol	Joback Method
hfus	18.75	kJ/mol	Joback Method
hvap	58.42	kJ/mol	Joback Method
log10ws	-1.98		Crippen Method
logp	2.025		Crippen Method
mcvol	120.340	ml/mol	McGowan Method
pc	4098.62	kPa	Joback Method
tb	548.45	K	Joback Method
tc	737.20	K	Joback Method
tf	312.93	K	Joback Method
vc	0.453	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.40	J/molxK	548.45	Joback Method
cpg	296.46	J/molxK	705.74	Joback Method
cpg	289.46	J/molxK	674.29	Joback Method
cpg	282.08	J/molxK	642.83	Joback Method

cpg	274.29	J/molxK	611.37	Joback Method
cpg	266.07	J/molxK	579.91	Joback Method
cpg	303.08	J/molxK	737.20	Joback Method
dvisc	0.0001610	Paxs	548.45	Joback Method
dvisc	0.0002512	Paxs	509.20	Joback Method
dvisc	0.0004220	Paxs	469.94	Joback Method
dvisc	0.0007795	Paxs	430.69	Joback Method
dvisc	0.0016283	Paxs	391.44	Joback Method
dvisc	0.0040083	Paxs	352.18	Joback Method
dvisc	0.0123694	Paxs	312.93	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	410.20	K	2.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C616057&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C616057&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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