

Cyclopropanamine, 2-phenyl-, trans-(+)-

Other names:	(+)-Tranylcypromine (+)-trans-2-Phenylcyclopropylamine (1R,2S)-rel-2-phenylcyclopropanamine (1S-trans)-2-Phenylcyclopropanamine 2-Phenyl-1-aminocyclopropane, trans- Cyclopropylamine, 2-phenyl-, (1S-trans)- Cyclopropylamine, 2-phenyl-, trans- D-Tranylcypromine dl-tranylcypromine trans-2-phenylcyclopropylamine
Inchi:	InChI=1S/C9H11N/c10-9-6-8(9)7-4-2-1-3-5-7/h1-5,8-9H,6,10H2
InchiKey:	AELCINSCMGFISI-UHFFFAOYSA-N
Formula:	C9H11N
SMILES:	NC1CC1c1ccccc1
Mol. weight [g/mol]:	133.19
CAS:	3721-28-6

Physical Properties

Property code	Value	Unit	Source
gf	256.80	kJ/mol	Joback Method
hf	93.69	kJ/mol	Joback Method
hfus	17.51	kJ/mol	Joback Method
hvap	48.15	kJ/mol	Joback Method
log10ws	-2.10		Crippen Method
logp	1.501		Crippen Method
mcvol	113.030	ml/mol	McGowan Method
pc	3920.94	kPa	Joback Method
rinpol	1240.30		NIST Webbook
tb	506.60	K	Joback Method
tc	743.89	K	Joback Method
tf	314.57	K	Joback Method
vc	0.416	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	324.78	J/mol×K	704.34	Joback Method
cpg	257.94	J/mol×K	506.60	Joback Method
cpg	273.48	J/mol×K	546.15	Joback Method
cpg	287.86	J/mol×K	585.70	Joback Method
cpg	301.16	J/mol×K	625.25	Joback Method
cpg	313.44	J/mol×K	664.79	Joback Method
cpg	335.27	J/mol×K	743.89	Joback Method
pvap	0.03	kPa	298.15	The Vaporization Enthalpies and Vapor Pressures of Some Primary Amines of Pharmaceutical Importance by Correlation Gas Chromatography McGowan Method:

Sources

The Vaporization Enthalpies and Vapor Pressures of Some Primary Amines of Pharmaceutical Importance by Correlation Gas Chromatography: McGowan Method:

<https://www.doi.org/10.1021/je400498a>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

Pharmaceutical Importance by Correlation Gas Chromatography:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3721286&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume

pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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