

# Fumaric acid, decyl 3-methylbut-2-yl ester

<b>Inchi:</b>	InChI=1S/C19H34O4/c1-5-6-7-8-9-10-11-12-15-22-18(20)13-14-19(21)23-17(4)16(2)3/h1
<b>InchiKey:</b>	RCQWGMWJUMGROF-BUHFOSPRSA-N
<b>Formula:</b>	C19H34O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C=CC(=O)OC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	326.47

## Physical Properties

Property code	Value	Unit	Source
gf	-283.40	kJ/mol	Joback Method
hf	-818.43	kJ/mol	Joback Method
hfus	43.70	kJ/mol	Joback Method
hvap	75.38	kJ/mol	Joback Method
log10ws	-5.22		Crippen Method
logp	4.814		Crippen Method
mcvol	289.150	ml/mol	McGowan Method
pc	1203.96	kPa	Joback Method
rinpola	2179.00		NIST Webbook
rinpola	2179.00		NIST Webbook
tb	789.98	K	Joback Method
tc	976.37	K	Joback Method
tf	413.13	K	Joback Method
vc	1.115	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	885.08	J/mol×K	789.98	Joback Method
cpg	963.08	J/mol×K	945.31	Joback Method
cpg	949.36	J/mol×K	914.24	Joback Method
cpg	934.72	J/mol×K	883.18	Joback Method
cpg	919.15	J/mol×K	852.11	Joback Method
cpg	902.61	J/mol×K	821.05	Joback Method
cpg	975.91	J/mol×K	976.37	Joback Method
dvisc	0.0000451	Paxs	789.98	Joback Method

dvisc	0.0000618	Paxs	727.17	Joback Method
dvisc	0.0000900	Paxs	664.36	Joback Method
dvisc	0.0001417	Paxs	601.56	Joback Method
dvisc	0.0002480	Paxs	538.75	Joback Method
dvisc	0.0005031	Paxs	475.94	Joback Method
dvisc	0.0012655	Paxs	413.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348082&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348082&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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