

# 1,1'-Biphenyl, 3-bromo-

<b>Other names:</b>	Biphenyl, 3-bromo- m-Bromobiphenyl 3-Bromobiphenyl 3-Bromo diphenyl
<b>Inchi:</b>	InChI=1S/C12H9Br/c13-12-8-4-7-11(9-12)10-5-2-1-3-6-10/h1-9H
<b>InchiKey:</b>	USYQKCQEVBFRP-UHFFFAOYSA-N
<b>Formula:</b>	C12H9Br
<b>SMILES:</b>	<chem>Brc1cccc(-c2ccccc2)c1</chem>
<b>Mol. weight [g/mol]:</b>	233.10
<b>CAS:</b>	2113-57-7

## Physical Properties

Property code	Value	Unit	Source
gf	279.67	kJ/mol	Joback Method
hf	196.91	kJ/mol	Joback Method
hfus	19.81	kJ/mol	Joback Method
hvap	53.95	kJ/mol	Joback Method
log10ws	-5.24		Crippen Method
logp	4.116		Crippen Method
mcvol	149.920	ml/mol	McGowan Method
pc	3655.35	kPa	Joback Method
tb	573.00 ± 1.00	K	NIST Webbook
tb	573.20	K	NIST Webbook
tc	864.14	K	Joback Method
tf	350.16	K	Joback Method
vc	0.553	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	315.92	J/mol×K	598.46	Joback Method
cpg	375.61	J/mol×K	819.86	Joback Method
cpg	365.83	J/mol×K	775.58	Joback Method
cpg	355.07	J/mol×K	731.30	Joback Method

cpg	343.23	J/molxK	687.02	Joback Method
cpg	330.21	J/molxK	642.74	Joback Method
cpg	384.51	J/molxK	864.14	Joback Method
dvisc	0.0002114	Paxs	598.46	Joback Method
dvisc	0.0002624	Paxs	557.08	Joback Method
dvisc	0.0003371	Paxs	515.69	Joback Method
dvisc	0.0004526	Paxs	474.31	Joback Method
dvisc	0.0006428	Paxs	432.93	Joback Method
dvisc	0.0009832	Paxs	391.54	Joback Method
dvisc	0.0016627	Paxs	350.16	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	376.70	K	0.03	NIST Webbook
tbrp	384.00	K	0.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2113577&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2113577&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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