

# 3-Phenylpropionic acid, 3,4-dichlorophenyl ester

Inchi:	InChI=1S/C15H12Cl2O2/c16-13-8-7-12(10-14(13)17)19-15(18)9-6-11-4-2-1-3-5-11/h1-5,
InchiKey:	FQOCOCXQBBVPNZ-UHFFFAOYSA-N
Formula:	C15H12Cl2O2
SMILES:	O=C(CCc1ccccc1)Oc1ccc(Cl)c(Cl)c1
Mol. weight [g/mol]:	295.16

## Physical Properties

Property code	Value	Unit	Source
gf	23.20	kJ/mol	Joback Method
hf	-179.09	kJ/mol	Joback Method
hfus	33.09	kJ/mol	Joback Method
hvap	72.79	kJ/mol	Joback Method
log10ws	-5.19		Crippen Method
logp	4.532		Crippen Method
mvol	206.610	ml/mol	McGowan Method
pc	2370.28	kPa	Joback Method
rinpol	2177.00		NIST Webbook
tb	757.07	K	Joback Method
tc	1001.10	K	Joback Method
tf	468.69	K	Joback Method
vc	0.781	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	513.29	J/molxK	757.07	Joback Method
cpg	566.61	J/molxK	960.43	Joback Method
cpg	557.99	J/molxK	919.75	Joback Method
cpg	548.39	J/molxK	879.08	Joback Method
cpg	537.77	J/molxK	838.41	Joback Method
cpg	526.09	J/molxK	797.74	Joback Method
cpg	574.32	J/molxK	1001.10	Joback Method
dvisc	0.0001143	Paxs	757.07	Joback Method
dvisc	0.0001416	Paxs	709.01	Joback Method

dvisc	0.0001811	Paxs	660.94	Joback Method
dvisc	0.0002408	Paxs	612.88	Joback Method
dvisc	0.0003359	Paxs	564.82	Joback Method
dvisc	0.0004986	Paxs	516.75	Joback Method
dvisc	0.0008024	Paxs	468.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307785&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307785&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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