

# Benzene, 1-(chloromethyl)-4-nitro-

<b>Other names:</b>	Toluene, «alpha»-chloro-p-nitro- «alpha»-Chloro-p-nitrotoluene p-Nitrobenzyl chloride 1-(Chloromethyl)-4-nitrobenzene 4-(Chloromethyl)nitrobenzene 4-Nitrobenzyl chloride p-(Chloromethyl)nitrobenzene Toluene, alpha-chloro-p-nitro NSC 9803 «alpha»-chloro-4-nitrotoluene
<b>Inchi:</b>	InChI=1S/C7H6ClNO2/c8-5-6-1-3-7(4-2-6)9(10)11/h1-4H,5H2
<b>InchiKey:</b>	KGCNHWXDPDPSBV-UHFFFAOYSA-N
<b>Formula:</b>	C7H6ClNO2
<b>SMILES:</b>	O=[N+](O-)c1ccc(CCl)cc1
<b>Mol. weight [g/mol]:</b>	171.58
<b>CAS:</b>	100-14-1

## Physical Properties

Property code	Value	Unit	Source
gf	134.46	kJ/mol	Joback Method
hf	10.75	kJ/mol	Joback Method
hfus	23.10	kJ/mol	Joback Method
hvap	55.09	kJ/mol	Joback Method
log10ws	-3.15		Crippen Method
logp	2.334		Crippen Method
mcvol	115.390	ml/mol	McGowan Method
pc	3911.14	kPa	Joback Method
tb	580.49	K	Joback Method
tc	834.39	K	Joback Method
tf	381.12	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	243.16	J/mol×K	580.49	Joback Method
cpg	253.38	J/mol×K	622.81	Joback Method
cpg	262.77	J/mol×K	665.12	Joback Method
cpg	271.39	J/mol×K	707.44	Joback Method
cpg	279.27	J/mol×K	749.76	Joback Method
cpg	286.47	J/mol×K	792.08	Joback Method
cpg	293.02	J/mol×K	834.39	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C100141&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C100141&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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