

# Aluminum, chlorodiethyl-

<b>Other names:</b>	Chlorodiethylaluminum Diethylaluminium chloride Diethylaluminum chloride Diethylaluminum monochloride Diethylchloroaluminum UN 1101
<b>Inchi:</b>	InChI=1S/2C2H5.Al.ClH/c2*1-2;;/h2*1H2,2H3;;1H/q;;+1;/p-1
<b>InchiKey:</b>	YNLAOSYQHBDIKW-UHFFFAOYSA-M
<b>Formula:</b>	C4H10AlCl
<b>SMILES:</b>	CC[AlH3](Cl)CC
<b>Mol. weight [g/mol]:</b>	120.56
<b>CAS:</b>	96-10-6

## Physical Properties

Property code	Value	Unit	Source
chl	-3627.50 ± 2.90	kJ/mol	NIST Webbook
hfl	-382.20 ± 3.50	kJ/mol	NIST Webbook
hfl	-423.10 ± 3.60	kJ/mol	NIST Webbook
hfl	-237.80 ± 3.00	kJ/mol	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	50.50	kJ/mol	298.00	NIST Webbook
hvapt	53.90	kJ/mol	373.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$

Coeff. A	1.69624e+01
Coeff. B	-5.72084e+03
Coeff. C	-1.77000e+01
Temperature range (K), min.	360.78
Temperature range (K), max.	508.72

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C96106&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>pvap:</b>	Vapor pressure

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