

# Glutaric acid, ethyl 2-methyl-4-chlorophenyl ester

Inchi:	InChI=1S/C14H17ClO4/c1-3-18-13(16)5-4-6-14(17)19-12-8-7-11(15)9-10(12)2/h7-9H,3-6
InchiKey:	JAODAZGBKOFEAD-UHFFFAOYSA-N
Formula:	C14H17ClO4
SMILES:	CCOC(=O)CCCC(=O)Oc1ccc(Cl)cc1C
Mol. weight [g/mol]:	284.74

## Physical Properties

Property code	Value	Unit	Source
gf	-319.62	kJ/mol	Joback Method
hf	-624.04	kJ/mol	Joback Method
hfus	35.05	kJ/mol	Joback Method
hvap	73.05	kJ/mol	Joback Method
log10ws	-3.90		Crippen Method
logp	3.287		Crippen Method
mvol	211.480	ml/mol	McGowan Method
pc	2054.89	kPa	Joback Method
rinpol	2074.00		NIST Webbook
tb	746.37	K	Joback Method
tc	956.53	K	Joback Method
tf	473.24	K	Joback Method
vc	0.808	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	564.84	J/molxK	746.37	Joback Method
cpg	621.94	J/molxK	921.51	Joback Method
cpg	612.31	J/molxK	886.48	Joback Method
cpg	601.78	J/molxK	851.45	Joback Method
cpg	590.37	J/molxK	816.42	Joback Method
cpg	578.06	J/molxK	781.40	Joback Method
cpg	630.68	J/molxK	956.53	Joback Method
dvisc	0.0001079	Paxs	746.37	Joback Method
dvisc	0.0001337	Paxs	700.85	Joback Method

dvisc	0.0001708	Paxs	655.33	Joback Method
dvisc	0.0002262	Paxs	609.81	Joback Method
dvisc	0.0003135	Paxs	564.28	Joback Method
dvisc	0.0004601	Paxs	518.76	Joback Method
dvisc	0.0007271	Paxs	473.24	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359011&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359011&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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