

# Fumaric acid, decyl 2,2,2-trichloroethyl ester

<b>Inchi:</b>	InChI=1S/C16H25Cl3O4/c1-2-3-4-5-6-7-8-9-12-22-14(20)10-11-15(21)23-13-16(17,18)19
<b>InchiKey:</b>	KZHVAYZESXTNDB-ZHACJKMWSA-N
<b>Formula:</b>	C16H25Cl3O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)C=CC(=O)OCC(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	387.73

## Physical Properties

Property code	Value	Unit	Source
gf	-336.73	kJ/mol	Joback Method
hf	-801.92	kJ/mol	Joback Method
hfus	48.15	kJ/mol	Joback Method
hvap	81.34	kJ/mol	Joback Method
log10ws	-5.66		Crippen Method
logp	5.140		Crippen Method
mcvol	283.600	ml/mol	McGowan Method
pc	1369.71	kPa	Joback Method
rinqol	2380.00		NIST Webbook
tb	831.28	K	Joback Method
tc	1031.86	K	Joback Method
tf	501.50	K	Joback Method
vc	1.095	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	796.28	J/molxK	831.28	Joback Method
cpg	855.06	J/molxK	998.43	Joback Method
cpg	844.87	J/molxK	965.00	Joback Method
cpg	833.95	J/molxK	931.57	Joback Method
cpg	822.24	J/molxK	898.14	Joback Method
cpg	809.70	J/molxK	864.71	Joback Method
cpg	864.55	J/molxK	1031.86	Joback Method
dvisc	0.0000415	Paxs	831.28	Joback Method
dvisc	0.0000548	Paxs	776.32	Joback Method

dvisc	0.0000757	Paxs	721.35	Joback Method
dvisc	0.0001102	Paxs	666.39	Joback Method
dvisc	0.0001716	Paxs	611.43	Joback Method
dvisc	0.0002918	Paxs	556.46	Joback Method
dvisc	0.0005571	Paxs	501.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348507&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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