

# Norbornene, 5-bromo-6-(trichloromethyl)

<b>Inchi:</b>	InChI=1S/C8H8BrCl3/c9-7-5-2-1-4(3-5)6(7)8(10,11)12/h1-2,4-7H,3H2
<b>InchiKey:</b>	CCIWXXGOEVIJVB-B-UHFFFAOYSA-N
<b>Formula:</b>	C8H8BrCl3
<b>SMILES:</b>	C1C(Cl)(Cl)C1C2C=CC(C2)C1Br
<b>Mol. weight [g/mol]:</b>	290.41

## Physical Properties

Property code	Value	Unit	Source
gf	121.79	kJ/mol	Joback Method
hf	-81.55	kJ/mol	Joback Method
hfus	24.47	kJ/mol	Joback Method
hvap	51.37	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.942		Crippen Method
mvol	151.780	ml/mol	McGowan Method
pc	3250.43	kPa	Joback Method
rinpol	1574.00		NIST Webbook
tb	565.23	K	Joback Method
tc	818.92	K	Joback Method
tf	356.54	K	Joback Method
vc	0.572	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.68	J/molxK	565.23	Joback Method
cpg	330.17	J/molxK	607.51	Joback Method
cpg	342.30	J/molxK	649.79	Joback Method
cpg	353.23	J/molxK	692.08	Joback Method
cpg	363.10	J/molxK	734.36	Joback Method
cpg	372.08	J/molxK	776.64	Joback Method
cpg	380.31	J/molxK	818.92	Joback Method
dvisc	0.0025890	Paxs	356.54	Joback Method
dvisc	0.0021780	Paxs	391.32	Joback Method

dvisc	0.0018846	Paxs	426.10	Joback Method
dvisc	0.0016668	Paxs	460.88	Joback Method
dvisc	0.0014997	Paxs	495.67	Joback Method
dvisc	0.0013683	Paxs	530.45	Joback Method
dvisc	0.0012625	Paxs	565.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R515454&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R515454&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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