

# 2-[2-(2-Ethoxyethoxy)ethyl] 2,2,3,3,3-pentafluoropropanoate

**Other names:** Triethylene glycol monoethyl ether, pentafluoropropionate

3,6,9-Trioxaundec-1-yl pentafluoropropionate

**Inchi:** InChI=1S/C11H17F5O5/c1-2-18-3-4-19-5-6-20-7-8-21-9(17)10(12,13)11(14,15)16/h2-8H

**InchiKey:** AGLKEEOFOBNSPE-UHFFFAOYSA-N

**Formula:** C11H17F5O5

**SMILES:** CCOCCOCCOCCOC(=O)C(F)(F)C(F)(F)F

**Mol. weight [g/mol]:** 324.24

## Physical Properties

Property code	Value	Unit	Source
gf	-1475.55	kJ/mol	Joback Method
hf	-1909.88	kJ/mol	Joback Method
hfus	31.17	kJ/mol	Joback Method
hvap	49.79	kJ/mol	Joback Method
log10ws	-1.53		Crippen Method
logp	1.797		Crippen Method
mcvol	199.750	ml/mol	McGowan Method
pc	1644.42	kPa	Joback Method
rinpol	1293.30		NIST Webbook
tb	584.52	K	Joback Method
tc	739.37	K	Joback Method
tf	360.37	K	Joback Method
vc	0.797	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.22	J/molxK	584.52	Joback Method
cpg	556.23	J/molxK	610.33	Joback Method
cpg	568.68	J/molxK	636.14	Joback Method
cpg	580.58	J/molxK	661.95	Joback Method
cpg	591.92	J/molxK	687.76	Joback Method
cpg	602.72	J/molxK	713.56	Joback Method
cpg	612.99	J/molxK	739.37	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U351985&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U351985&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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