

# (3-Methylphenyl) methanol, neopentyl ether

<b>Inchi:</b>	InChI=1S/C13H20O/c1-11-6-5-7-12(8-11)9-14-10-13(2,3)4/h5-8H,9-10H2,1-4H3
<b>InchiKey:</b>	AGBLWVRAXHXNJN-UHFFFAOYSA-N
<b>Formula:</b>	C13H20O
<b>SMILES:</b>	Cc1cccc(COCC(C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	192.30

## Physical Properties

Property code	Value	Unit	Source
gf	59.20	kJ/mol	Joback Method
hf	-227.56	kJ/mol	Joback Method
hfus	16.85	kJ/mol	Joback Method
hvap	48.58	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.558		Crippen Method
mcvol	176.140	ml/mol	McGowan Method
pc	2167.36	kPa	Joback Method
rinpol	1325.00		NIST Webbook
tb	547.69	K	Joback Method
tc	755.69	K	Joback Method
tf	299.86	K	Joback Method
vc	0.662	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	423.50	J/molxK	547.69	Joback Method
cpg	441.42	J/molxK	582.36	Joback Method
cpg	458.32	J/molxK	617.02	Joback Method
cpg	474.24	J/molxK	651.69	Joback Method
cpg	489.20	J/molxK	686.36	Joback Method
cpg	503.27	J/molxK	721.02	Joback Method
cpg	516.46	J/molxK	755.69	Joback Method
dvisc	0.0024476	Paxs	299.86	Joback Method
dvisc	0.0011461	Paxs	341.17	Joback Method

dvisc	0.0006322	Paxs	382.47	Joback Method
dvisc	0.0003916	Paxs	423.78	Joback Method
dvisc	0.0002641	Paxs	465.08	Joback Method
dvisc	0.0001900	Paxs	506.39	Joback Method
dvisc	0.0001436	Paxs	547.69	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374441&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374441&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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