

# 17-Norkaur-15-ene, 13-methyl-, (8«beta»,13«beta»)-

**Other names:** 17-Nor-8«beta»,13«beta»-kaur-15-ene, 13-methyl-

(+)-Beyerene  
(+)-Hibaene  
(+)-Stachene  
ent-Beyer-15-ene  
Beyerene  
Hibaene, (+)-  
Stachen  
Stachene  
Stachene, (+)-  
Stach-15-ene, (+)-  
Hibaene

**Inchi:** InChI=1S/C20H32/c1-17(2)8-5-9-19(4)15(17)7-11-20-13-12-18(3,14-20)10-6-16(19)20/h

**InchiKey:** GXMKKDDGINQVBE-UHFFFAOYSA-N

**Formula:** C20H32

**SMILES:** CC12C=CC3(CCC4C(C)(C)CCCC4(C)C3CC1)C2

**Mol. weight [g/mol]:** 272.47

**CAS:** 3564-54-3

## Physical Properties

Property code	Value	Unit	Source
gf	304.70	kJ/mol	Joback Method
hf	-111.51	kJ/mol	Joback Method
hfus	9.87	kJ/mol	Joback Method
hvap	55.52	kJ/mol	Joback Method
log10ws	-6.18		Crippen Method
logp	5.975		Crippen Method
mcvol	244.920	ml/mol	McGowan Method
pc	1783.35	kPa	Joback Method
rinpol	1924.00		NIST Webbook
rinpol	1962.00		NIST Webbook
rinpol	1913.00		NIST Webbook
rinpol	1913.00		NIST Webbook
rinpol	1913.00		NIST Webbook
rinpol	1951.00		NIST Webbook
rinpol	1924.00		NIST Webbook
rinpol	1941.00		NIST Webbook

ripol	2200.00		NIST Webbook
ripol	2241.00		NIST Webbook
ripol	2200.00		NIST Webbook
tb	691.82	K	Joback Method
tc	943.08	K	Joback Method
tf	460.72	K	Joback Method
vc	0.927	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	765.29	J/mol×K	691.82	Joback Method
cpg	793.16	J/mol×K	733.70	Joback Method
cpg	820.50	J/mol×K	775.57	Joback Method
cpg	848.06	J/mol×K	817.45	Joback Method
cpg	876.55	J/mol×K	859.33	Joback Method
cpg	906.69	J/mol×K	901.20	Joback Method
cpg	939.22	J/mol×K	943.08	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3564543&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3564543&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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