

d-Mannose

Other names: Carubiose; Mannose; Mannose, d-; Seminose; d(+)-Mannose.

InChI: InChI=1S/C6H12O6/c7-1-3(9)5(11)6(12)4(10)2-8/h1,3-6,8-12H,2H2/t3-,4-,5-,6-/m1/s1

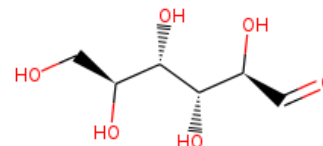
InChI Key: GZCGUPFRVQAUEE-KVTDHHQDSA-N

Formula: C6H12O6

SMILES: O=CC(O)C(O)C(O)C(O)CO

Molecular Weight: 180.16

CAS: 3458-28-4



Physical Properties

Property	Value	Unit	Source
$\Delta_c H^\circ_{\text{solid}}$	-2812.67 ± 0.93	kJ/mol	NIST Webbook
$\Delta_c H^\circ_{\text{solid}}$	-2813.00 ± 3.40	kJ/mol	NIST Webbook
$\Delta_f G^\circ$	-793.74	kJ/mol	Joback Method
$\Delta_f H^\circ_{\text{gas}}$	-1035.02	kJ/mol	Joback Method
$\Delta_f H^\circ_{\text{solid}}$	-1263.40 ± 1.20	kJ/mol	NIST Webbook
$\Delta_{\text{fus}} H^\circ$	19.93	kJ/mol	Joback Method
$\Delta_{\text{vap}} H^\circ$	117.51	kJ/mol	Joback Method
$\log P_{\text{oct/wat}}$	-3.379		Crippen Method
P_c	6631.37	kPa	Joback Method
T_{boil}	844.48	K	Joback Method
T_c	1034.02	K	Joback Method
T_{fus}	443.48	K	Joback Method
V_c	0.460	m ³ /kg-mol	Joback Method

Temperature Dependent Properties

Property	Value	Unit	Temperature (K)	Source
$C_{p,gas}$	390.04	J/mol×K	844.48	Joback Method
$C_{p,solid}$	214.20	J/mol×K	300.0	NIST Webbook
$C_{p,solid}$	216.00	J/mol×K	303.0	NIST Webbook
η	0.0000000	Paxs	844.48	Joback Method

Sources

Joback Method: https://en.wikipedia.org/wiki/Joback_method

NIST Webbook: [http://webbook.nist.gov/cgi/inchi/InChI=1S/C6H12O6/c7-1-3\(9\)5\(11\)6\(12\)4\(10\)2-8/h1,3-6,8-12H,2H2/t3-,4-,5-,6-/m1/s1](http://webbook.nist.gov/cgi/inchi/InChI=1S/C6H12O6/c7-1-3(9)5(11)6(12)4(10)2-8/h1,3-6,8-12H,2H2/t3-,4-,5-,6-/m1/s1)

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Legend

$\Delta_c H^\circ_{solid}$: Standard solid enthalpy of combustion (kJ/mol).

$C_{p,gas}$: Ideal gas heat capacity (J/mol×K).

$C_{p,solid}$: Solid phase heat capacity (J/mol×K).

η : Dynamic viscosity (Paxs).

$\Delta_f G^\circ$: Standard Gibbs free energy of formation (kJ/mol).

$\Delta_f H^\circ_{gas}$: Enthalpy of formation at standard conditions (kJ/mol).

$\Delta_f H^\circ_{solid}$: Solid phase enthalpy of formation at standard conditions (kJ/mol).

$\Delta_{fus} H^\circ$: Enthalpy of fusion at standard conditions (kJ/mol).

$\Delta_{vap} H^\circ$: Enthalpy of vaporization at standard conditions (kJ/mol).

$\log P_{oct/wat}$: Octanol/Water partition coefficient .

P_c : Critical Pressure (kPa).

T_{boil} : Normal Boiling Point Temperature (K).

T_c : Critical Temperature (K).

T_{fus} : Normal melting (fusion) point (K).

V_c : Critical Volume (m³/kg-mol).

Latest version available from:

<https://www.cheméo.com/cid/35-097-6/d-Mannose>

Generated by Cheméo on Sat, 16 Feb 2019 01:40:09 +0000.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.