

# p-Methylcinnamic acid

<b>Other names:</b>	4-Methylcinnamic acid 2-Propenoic acid, 3-(4-methylphenyl)- Cinnamic acid, p-methyl- trans-p-methylcinnamic acid
<b>Inchi:</b>	InChI=1S/C10H10O2/c1-8-2-4-9(5-3-8)6-7-10(11)12/h2-7H,1H3,(H,11,12)/b7-6+
<b>InchiKey:</b>	RURHILYUWQEGOS-VOTSOKGWSA-N
<b>Formula:</b>	C10H10O2
<b>SMILES:</b>	<chem>Cc1ccc(C=CC(=O)O)cc1</chem>
<b>Mol. weight [g/mol]:</b>	162.19
<b>CAS:</b>	1866-39-3

## Physical Properties

Property code	Value	Unit	Source
gf	-49.42	kJ/mol	Joback Method
hf	-172.26	kJ/mol	Joback Method
hfus	21.20	kJ/mol	Joback Method
hvap	64.18	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	2.093		Crippen Method
mcvol	131.140	ml/mol	McGowan Method
pc	3650.93	kPa	Joback Method
tb	610.07	K	Joback Method
tc	819.56	K	Joback Method
tf	347.07	K	Joback Method
vc	0.492	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	305.72	J/mol×K	610.07	Joback Method
cpg	351.51	J/mol×K	784.64	Joback Method
cpg	343.54	J/mol×K	749.73	Joback Method
cpg	335.02	J/mol×K	714.81	Joback Method
cpg	325.90	J/mol×K	679.90	Joback Method

cpg	316.15	J/mol×K	644.98	Joback Method
cpg	358.96	J/mol×K	819.56	Joback Method
dvisc	0.0000748	Paxs	610.07	Joback Method
dvisc	0.0001119	Paxs	566.24	Joback Method
dvisc	0.0001792	Paxs	522.40	Joback Method
dvisc	0.0003127	Paxs	478.57	Joback Method
dvisc	0.0006105	Paxs	434.74	Joback Method
dvisc	0.0013852	Paxs	390.90	Joback Method
dvisc	0.0038653	Paxs	347.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1866393&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1866393&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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