

# Acetic acid, phenoxy-

<b>Other names:</b>	2-Phenoxyacetic acid Acetic acid, 2-phenoxy- Acide phenoxyacetique Glycolic acid, phenyl ether Glycollic acid phenyl ether NSC 9810 POA Phenoxyacetic acid Phenoxyethanoic acid o-Phenyglycolic acid
<b>Inchi:</b>	InChI=1S/C8H8O3/c9-8(10)6-11-7-4-2-1-3-5-7/h1-5H,6H2,(H,9,10)
<b>InchiKey:</b>	LCPDWSOZIOUXRV-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O3
<b>SMILES:</b>	O=C(O)COc1ccccc1
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	122-59-8

## Physical Properties

Property code	Value	Unit	Source
gf	-241.85	kJ/mol	Joback Method
hf	-368.95	kJ/mol	Joback Method
hfus	17.39	kJ/mol	Joback Method
hvap	61.51	kJ/mol	Joback Method
log10ws	-1.10		Aqueous Solubility Prediction Method
logp	1.150		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	4345.39	kPa	Joback Method
tb	577.59	K	Joback Method
tc	781.86	K	Joback Method
tf	372.65	K	Aqueous Solubility Prediction Method
vc	0.418	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.57	J/molxK	577.59	Joback Method
cpg	268.05	J/molxK	611.63	Joback Method
cpg	276.96	J/molxK	645.68	Joback Method
cpg	285.33	J/molxK	679.72	Joback Method
cpg	293.16	J/molxK	713.77	Joback Method
cpg	300.47	J/molxK	747.81	Joback Method
cpg	307.28	J/molxK	781.86	Joback Method
dvisc	0.0046046	Paxs	339.32	Joback Method
dvisc	0.0017223	Paxs	379.03	Joback Method
dvisc	0.0007763	Paxs	418.74	Joback Method
dvisc	0.0004017	Paxs	458.46	Joback Method
dvisc	0.0002309	Paxs	498.17	Joback Method
dvisc	0.0001440	Paxs	537.88	Joback Method
dvisc	0.0000958	Paxs	577.59	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C122598&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Solubilities of cinnamic acid, phenoxyacetic acid and 4-hydroxyphenylacetic acid in supercritical carbon dioxide:** <https://www.doi.org/10.1016/j.fluid.2008.09.009>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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