

# Adipic acid, monochloride 3,3-dimethylbut-2-yl ester

Inchi:	InChI=1S/C12H21ClO3/c1-9(12(2,3)4)16-11(15)8-6-5-7-10(13)14/h9H,5-8H2,1-4H3
InchiKey:	HYCAONTXXYAGCB-UHFFFAOYSA-N
Formula:	C12H21ClO3
SMILES:	CC(OC(=O)CCCCC(=O)Cl)C(C)(C)C
Mol. weight [g/mol]:	248.75

## Physical Properties

Property code	Value	Unit	Source
gf	-324.21	kJ/mol	Joback Method
hf	-678.16	kJ/mol	Joback Method
hfus	24.48	kJ/mol	Joback Method
hvap	60.91	kJ/mol	Joback Method
log10ws	-3.51		Crippen Method
logp	3.290		Crippen Method
mcvol	201.190	ml/mol	McGowan Method
pc	1938.95	kPa	Joback Method
rinpola	1557.00		NIST Webbook
tb	637.88	K	Joback Method
tc	832.11	K	Joback Method
tf	364.43	K	Joback Method
vc	0.769	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	521.95	J/molxK	637.88	Joback Method
cpg	536.76	J/molxK	670.25	Joback Method
cpg	550.74	J/molxK	702.62	Joback Method
cpg	563.92	J/molxK	734.99	Joback Method
cpg	576.32	J/molxK	767.36	Joback Method
cpg	587.97	J/molxK	799.73	Joback Method
cpg	598.91	J/molxK	832.11	Joback Method
dvisc	0.0027638	Paxs	364.43	Joback Method
dvisc	0.0012875	Paxs	410.00	Joback Method

dvisc	0.0006988	Paxs	455.58	Joback Method
dvisc	0.0004239	Paxs	501.15	Joback Method
dvisc	0.0002795	Paxs	546.73	Joback Method
dvisc	0.0001964	Paxs	592.30	Joback Method
dvisc	0.0001452	Paxs	637.88	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353685&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353685&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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