

# 2-Propenoic acid, 3-phenyl-, 1-methylethyl ester

<b>Other names:</b>	Cinnamic acid, isopropyl ester Isopropyl cinnamate Isopropylester kyseliny skoricove Isopropyl 3-phenylpropenoate 1-Methylethyl 3-phenylpropenoate Isopropyl 3-phenyl-2-propenoate
<b>Inchi:</b>	InChI=1S/C12H14O2/c1-10(2)14-12(13)9-8-11-6-4-3-5-7-11/h3-10H,1-2H3/b9-8+
<b>InchiKey:</b>	RGACABDFLVLVCT-CMDGGGOBGSA-N
<b>Formula:</b>	C12H14O2
<b>SMILES:</b>	CC(C)OC(=O)C=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	190.24
<b>CAS:</b>	7780-06-5

## Physical Properties

Property code	Value	Unit	Source
gf	6.43	kJ/mol	Joback Method
hf	-187.34	kJ/mol	Joback Method
hfus	20.34	kJ/mol	Joback Method
hvap	53.31	kJ/mol	Joback Method
log10ws	-2.94		Crippen Method
logp	2.651		Crippen Method
mcvol	159.320	ml/mol	McGowan Method
pc	2676.31	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1485.00		NIST Webbook
ripol	2097.00		NIST Webbook
ripol	2097.00		NIST Webbook
tb	580.65	K	Joback Method
tc	800.23	K	Joback Method
tf	303.50	K	Joback Method
vc	0.598	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	374.76	J/molxK	580.65	Joback Method
cpg	389.79	J/molxK	617.25	Joback Method
cpg	403.87	J/molxK	653.84	Joback Method
cpg	417.03	J/molxK	690.44	Joback Method
cpg	429.31	J/molxK	727.04	Joback Method
cpg	440.76	J/molxK	763.63	Joback Method
cpg	451.41	J/molxK	800.23	Joback Method
dvisc	0.0027204	Paxs	303.50	Joback Method
dvisc	0.0012128	Paxs	349.69	Joback Method
dvisc	0.0006529	Paxs	395.88	Joback Method
dvisc	0.0004000	Paxs	442.07	Joback Method
dvisc	0.0002689	Paxs	488.27	Joback Method
dvisc	0.0001936	Paxs	534.46	Joback Method
dvisc	0.0001469	Paxs	580.65	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7780065&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7780065&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/35-224-4/2-Propenoic-acid-3-phenyl-1-methylethyl-ester.pdf>

Generated by Cheméo on 2024-04-17 23:43:42.153052103 +0000 UTC m=+15686671.073629420.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.