

# 1-chlorobutyl trichloroacetate

<b>Other names:</b>	1-Butanol, 1-chloro, trichloroacetate
<b>Inchi:</b>	InChI=1S/C6H8Cl4O2/c1-2-3-4(7)12-5(11)6(8,9)10/h4H,2-3H2,1H3
<b>InchiKey:</b>	DCRNHUUZGUDKLN-UHFFFAOYSA-N
<b>Formula:</b>	C6H8Cl4O2
<b>SMILES:</b>	CCCC(Cl)OC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	253.94

## Physical Properties

Property code	Value	Unit	Source
gf	-281.60	kJ/mol	Joback Method
hf	-488.96	kJ/mol	Joback Method
hfus	19.93	kJ/mol	Joback Method
hvap	53.96	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.265		Crippen Method
mcvol	151.800	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
rinpol	1201.00		NIST Webbook
rinpol	1237.00		NIST Webbook
rinpol	1208.00		NIST Webbook
rinpol	1244.00		NIST Webbook
ripol	1647.00		NIST Webbook
ripol	1650.00		NIST Webbook
ripol	1657.00		NIST Webbook
ripol	1661.00		NIST Webbook
ripol	1678.00		NIST Webbook
tb	559.02	K	Joback Method
tc	776.12	K	Joback Method
tf	336.64	K	Joback Method
vc	0.575	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	298.23	J/molxK	559.02	Joback Method
cpg	307.26	J/molxK	595.20	Joback Method
cpg	315.65	J/molxK	631.39	Joback Method
cpg	323.42	J/molxK	667.57	Joback Method
cpg	330.62	J/molxK	703.76	Joback Method
cpg	337.27	J/molxK	739.94	Joback Method
cpg	343.39	J/molxK	776.12	Joback Method
dvisc	0.0032699	Paxs	336.64	Joback Method
dvisc	0.0017065	Paxs	373.70	Joback Method
dvisc	0.0010015	Paxs	410.77	Joback Method
dvisc	0.0006420	Paxs	447.83	Joback Method
dvisc	0.0004404	Paxs	484.89	Joback Method
dvisc	0.0003188	Paxs	521.96	Joback Method
dvisc	0.0002408	Paxs	559.02	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R111532&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R111532&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/35-269-5/1-chlorobutyl-trichloroacetate.pdf>

Generated by Cheméo on 2024-04-18 21:41:09.433493075 +0000 UTC m=+15765718.354070391.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.