

# p-Chlorophenylacetyl chloride

<b>Other names:</b>	4-Chlorophenacyl chloride 4-Chlorophenylacetyl chloride Benzeneacetyl chloride, 4-chloro-
<b>Inchi:</b>	InChI=1S/C8H6Cl2O/c9-7-3-1-6(2-4-7)5-8(10)11/h1-4H,5H2
<b>InchiKey:</b>	UMQUIRYNOVNYPA-UHFFFAOYSA-N
<b>Formula:</b>	C8H6Cl2O
<b>SMILES:</b>	O=C(Cl)Cc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	189.04
<b>CAS:</b>	25026-34-0

## Physical Properties

Property code	Value	Unit	Source
gf	-33.52	kJ/mol	Joback Method
hf	-127.45	kJ/mol	Joback Method
hfus	20.12	kJ/mol	Joback Method
hvap	51.86	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	2.648		Crippen Method
mcvol	125.870	ml/mol	McGowan Method
pc	3538.87	kPa	Joback Method
tb	542.83	K	Joback Method
tc	776.56	K	Joback Method
tf	328.63	K	Joback Method
vc	0.479	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	238.61	J/molxK	542.83	Joback Method
cpg	248.45	J/molxK	581.78	Joback Method
cpg	257.58	J/molxK	620.74	Joback Method
cpg	266.02	J/molxK	659.69	Joback Method
cpg	273.83	J/molxK	698.65	Joback Method
cpg	281.02	J/molxK	737.60	Joback Method

cpg	287.64	J/molxK	776.56	Joback Method
dvisc	0.0021405	Paxs	328.63	Joback Method
dvisc	0.0013147	Paxs	364.33	Joback Method
dvisc	0.0008809	Paxs	400.03	Joback Method
dvisc	0.0006303	Paxs	435.73	Joback Method
dvisc	0.0004744	Paxs	471.43	Joback Method
dvisc	0.0003716	Paxs	507.13	Joback Method
dvisc	0.0003006	Paxs	542.83	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	394.50 ± 1.50	K	2.70	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25026340&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25026340&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/35-440-4/p-Chlorophenylacetyl-chloride.pdf>

Generated by Cheméo on 2024-04-17 21:53:19.409631039 +0000 UTC m=+15680048.330208355.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.