

# Erythritol, tetrakis(trifluoroacetate), meso-

<b>Other names:</b>	Erythritol, tetrakis(trifluoroacetate) meso-Erythritol, tetrakis(trifluoroacetate)
<b>Inchi:</b>	InChI=1S/C12H6F12O8/c13-9(14,15)5(25)29-1-3(31-7(27)11(19,20)21)4(32-8(28)12(22,23)24)6(26)10(16,17)18
<b>InchiKey:</b>	SIWZLDBXSQMWPJ-UHFFFAOYSA-N
<b>Formula:</b>	C12H6F12O8
<b>SMILES:</b>	O=C(OCC(OC(=O)C(F)(F)F)C(COC(=O)C(F)(F)F)OC(=O)C(F)(F)F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	506.15
<b>CAS:</b>	27088-70-6

## Physical Properties

Property code	Value	Unit	Source
gf	-3216.76	kJ/mol	Joback Method
hf	-3669.09	kJ/mol	Joback Method
hfus	38.24	kJ/mol	Joback Method
hvap	63.17	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	2.146		Crippen Method
mcvol	230.940	ml/mol	McGowan Method
pc	1441.35	kPa	Joback Method
rinpola	992.10		NIST Webbook
tb	756.56	K	Joback Method
tc	927.46	K	Joback Method
tf	500.40	K	Joback Method
vc	0.964	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	708.26	J/molxK	756.56	Joback Method
cpg	716.89	J/molxK	785.04	Joback Method
cpg	724.78	J/molxK	813.53	Joback Method
cpg	731.96	J/molxK	842.01	Joback Method
cpg	738.46	J/molxK	870.49	Joback Method
cpg	744.31	J/molxK	898.97	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27088706&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27088706&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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