

# [1,1'-Biphenyl]-3-amine

<b>Other names:</b>	3-Biphenylamine 3-Aminobiphenyl m-Aminobiphenyl m-Phenylaniline 3-Phenylaniline
<b>Inchi:</b>	InChI=1S/C12H11N/c13-12-8-4-7-11(9-12)10-5-2-1-3-6-10/h1-9H,13H2
<b>InchiKey:</b>	MUNOBADFTHUUG-UHFFFAOYSA-N
<b>Formula:</b>	C12H11N
<b>SMILES:</b>	<chem>Nc1cccc(-c2ccccc2)c1</chem>
<b>Mol. weight [g/mol]:</b>	169.22
<b>CAS:</b>	2243-47-2

## Physical Properties

Property code	Value	Unit	Source
gf	331.80	kJ/mol	Joback Method
hf	204.37	kJ/mol	Joback Method
hfus	19.73	kJ/mol	Joback Method
hvap	58.16	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	2.936		Crippen Method
mcvol	142.400	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
rinpol	297.33		NIST Webbook
tb	604.83	K	Joback Method
tc	863.82	K	Joback Method
tf	304.30 ± 0.30	K	NIST Webbook
vc	0.520	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	335.82	J/mol×K	604.83	Joback Method
cpg	350.95	J/mol×K	647.99	Joback Method
cpg	364.81	J/mol×K	691.16	Joback Method

cpg	377.49	J/mol×K	734.32	Joback Method
cpg	389.06	J/mol×K	777.49	Joback Method
cpg	399.61	J/mol×K	820.65	Joback Method
cpg	409.23	J/mol×K	863.82	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	527.20	K	18.00	NIST Webbook
tbrp	450.20	K	2.40	NIST Webbook
tbrp	527.00	K	18.00	NIST Webbook
tbrp	450.50 ± 0.50	K	2.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2243472&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2243472&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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