

# 16«alpha»,17-dihydroxy-16,17-dihydroGA4-17-O-«permethylated»

InChI: InChI=1S/C32H50O12/c1-29-20(36-3)11-12-32(44-28(29)34)19-10-9-17-13-30(19,21(25)  
InChIKey: DYZOPYIERZRHQX-QULGNCHQSA-N

Formula: C32H50O12

SMILES: COCC1OC(OCC2(OC)CC34CC2CCC3C23CCC(OC)C(C)(C(=O)O2)C3C4C(=O)OC)C(C

Mol. weight [g/mol]: 626.73

## Physical Properties

Property code	Value	Unit	Source
gf	-824.83	kJ/mol	Joback Method
hf	-1971.61	kJ/mol	Joback Method
hfus	66.89	kJ/mol	Joback Method
hvap	119.38	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	2.141		Crippen Method
mcbvol	458.420	ml/mol	McGowan Method
pc	804.79	kPa	Joback Method
rinsol	3775.00		NIST Webbook
rinsol	3780.00		NIST Webbook
rinsol	3775.00		NIST Webbook
tb	1311.90	K	Joback Method
tc	1621.26	K	Joback Method
tf	951.25	K	Joback Method
vc	1.712	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2234.19	J/mol×K	1311.90	Joback Method
cpg	2324.27	J/mol×K	1363.46	Joback Method
cpg	2422.92	J/mol×K	1415.02	Joback Method
cpg	2531.13	J/mol×K	1466.58	Joback Method
cpg	2649.89	J/mol×K	1518.14	Joback Method
cpg	2780.19	J/mol×K	1569.70	Joback Method
cpg	2923.02	J/mol×K	1621.26	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R259882&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R259882&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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